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(54) Title: ABSORBENT ARTICLE

(57) Abstract: The invention relates to the use of a layer obtainable by a process comprising (a) forming a sprayable blend comprising one or more superabsorbent forming monomers superabsorbent polymer particles water, and one or more initiators, (b) applying said sprayable blend on a fibrous web; and (c) subjecting said fibrous web to conditions under which the superabsorbent forming monomer polymerizes, as a storage layer for aqueous fluids.

**Absorbent article****Description**

The invention relates to a storage layer in an absorbent article which is useful in hygiene products.

Superabsorbent polymers used in personal care products are designed to absorb body fluids, such as urine, blood, menses and the like. Generally, body fluids are containing malodorous components. Such odor develops in contact with air and bacteria for prolonged periods. Furthermore, absorbed exudates are converted to ammonia by urease produced by skin-florea, which can lead to skin irritations. Odor control materials have already been used in superabsorbent polymers. Odor control materials known in the art include zeolites, silica, activated carbon, chelants, antimicrobial agents, perfuming ingredients, masking agents, enzymes, peroxides, hydrogencarbonates, extracts of vegetables, etherical oils, bor derivatives, poly-alpha-aminoacids, imides, polyimides, pvp-iodine, chitosane, polyglycoside, and cyclophanes, cf. for example U.S. Patent 4,795,482, EP-B-0,297,543, EP-A-0 341 951, US-A-4 990 338 and US-A-5 035 892.

U.S. Patent 5,037,412 describes the use of odor control material which absorbs odor compounds of each level of pH. Materials for absorption of acidic odors are anorganical carbonates, bicarbonates, phosphates, biphosphates, sulfates, bisulfate or mixtures thereof with pH of higher than 7. These inorganic compounds are present in the odor control material from 40 bis 65 % of the whole mixture. Absorption of odors of pH of higher than 7 is done by materials of pH of lower than 7, e. g. ascorbic acid, stearic acid, citric acid, maleic acid, and polyacrylic acid. These compounds are added from 30 - 45 % to the mixture. Neutral odors are absorbed by activated carbon, silica, polystyrene derivatives, zeolites, molecular sieve and starch with contents up to 10 % of the whole mixture. The application of said dry mixture has the advantage of no special production or handling compared to fluid products (deodorants). The odor control material is incorporated in personal care products as a separat unit beside the superabsorbent material. Unfortunately, in such products the absorption capacity of the superabsorbent polymer is reduced.

U.S. Patent 4,795,482 and EP-B-0,297,543 propose molecular sieve of typus  $\text{SiO}_2/\text{Al}_2\text{O}_3$  of molar ratio from 35 to infinity, preferably from 200 bis 500, for use in odor control. At least 90 % of the structure has to be built from the tetrahedral structure of si-

lica. The average diameter of the pores should be at least 0.55 nm. Unfortunately structures disclosed in these patents show only small absorption capacities of lower than 10 % by weight (tested with water).

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WO-A-98/28478 describes layered structures of personal care products comprising hydrophilic fibers prepared and stabilized by addition of latices. Odor control material and superabsorbent material is added. Odor control compounds are for example disclosed 10 in U.S. Patent 3,093,546 (antimicrobial agents, e. g. halogenated phenylene), U.S. Patent 4,385,632 (copper derivatives, preferably copper acetate), U.S. Patent 4,525,410 (zeolite), U.S. Patent 5,037,412 (acidic compounds as ascorbic acid, stearic acid, citric acid, maleic acid, polyacrylic acid). Mixtures of silica: 15 zeolite : superabsorbent material are for instance used in the ratio 1:5:1 to 1:1:5.

EP-A-0,894,502 relates to the use of  $\alpha$ -cyclodextrin of particle size 12 to 800  $\mu\text{m}$  as odor control material in an absorbent article containing a fluid-absorbing material such as superabsorbent polymer particles or an absorbent foam. According to WO-A-00/66187 odor-controlling superabsorbent polymer particles contain an odor-controlling compound such as cyclodextrin, triclosan, an amphoteric surfactant, a water-insoluble phosphate or 25 mixtures thereof homogeneously distributed throughout the particle.

As already mentioned above, using odor-control materials and superabsorbents separately leads to an absorption profile both for 30 body fluid and odor which is far away from each compound used alone. Further such high loadings with odor control material results in a worse absorption profile of the superabsorbent polymers.

35 EP-A-0 341 951, US-A-4 990 338 and US-A-5 035 892 teach the addition of silica in preparation of antimicrobial polymers. Generally the addition of silica to superabsorbent particles results in covering the surface area of the particles and therefore influences only the properties of the surface of superabsorbent polymers and not the intrinsic absorption properties of the hydrogel particles. Thus the addition of silica-compounds leads to hydrophilization or hydrophobation of the superabsorbent particles 40 and influences at first the absorption rate.

45 Despite optimizing the superabsorbent material there still exists a phenomena called "gel-blocking". Gel blocking occurs when superabsorbent particles are wetted and the particles swell at the

surface so as to inhibit fluid transmission to the inner part of the particle or to other regions of the absorbent structure. Wetting of other regions takes place by a very slow diffusion process. As a consequence, acquisition of aqueous body fluids by the 5 absorbent structure is much slower than the rate, at which the fluids are discharged, especially in gush situations. Leakage takes place, before the superabsorbent particles are fully saturated and the fluids reach the unwetted regions.

10 Thus the gel-blocking phenomena necessitated the use of a matrix cellululosic fibers to serve as temporarily fluid storage reservoir. The aqueous body fluid is held in the pores of the fluff matrix. The superabsorbent particles are dispersed in the matrix and thus separated from one another. The superabsorbent particles 15 then absorbs the fluid and deters the fluff matrix. That is, the fluff serves to absorb and hold relatively large quantities of quickly applied body fluids until the superabsorbent material quantitatively absorb the body fluid. From this point of view it is impossible to create continuous layers of superabsorbent hy- 20 drogels.

Accordingly, a number of different structures have been proposed to overcome the problems associated with incorporating superabsorbent materials in personal care products.

25 U.S. Patent 4,699,619 describes a multi-layer absorbent composite having a first relatively low density layer and a second relatively high density layer underlying at least a portion of the first layer. A high absorbency material (superabsorbent) is des- 30 cribed as being located between said first and second layers.

U.S. Patent 4,102,340 relates to a disposable article with a particulate hydrophilic polymer in an absorbent bed. Described is an absorbent pad comprising a fibrous structure having an intermedi- 35 ate densified layer and a layer of highly porous, loosely compacted bed on both sides of the densified layer. One of the bed layers is described as including a particulate, water-insoluble but water-swellable polymeric absorbent.

40 U.S. Patent 4,269,188 is directed to a disposable diaper. Disclosed is a disposable diaper including an absorbent material comprising a laminate wherein a water-absorbent polymer powder is fixed between two sheets of paper. A fluff pulp layer is located on both sides of the laminate containing a water-absorbing poly- 45 mer.

While the structures described in the referenced patents have often proven beneficial, they have not completely solved the problems associated with the use of superabsorbent materials. Thus by using conventional methods, the wood pulp fluff serves to quickly absorb and hold surges (relatively large quantities applied relatively quickly) of body fluid until the high-absorbency material can absorb it. Then the fluff matrix is dewatered by high-absorbency material. Because of gel-blocking, the ratio of superabsorbent to fiber remains too low for the intended use of absorbing large quantities of body fluids. Furthermore, the problem of separation of the incorporated and not affixed particulates by wearing the personal care product remains. Up to now there is no proposed absorbent structure that overcome the problem of incorporation higher amounts of superabsorbent materials without changing the absorption profile.

It is desired to provide an absorbent structure, which contains immobilized superabsorbent particles affixed to each other or onto a substrate. It is further desired to create an absorbent layer having a relatively high concentration of high-absorbency material but which absorbent structure is capable of absorbing and storing body fluids applied thereto.

A lot of work was done to immobilize superabsorbent particles. One possibility is the addition of large quantities of liquid polyhydroxy compounds that act as an adhesive to hold the particles together or attach the particles to a substrate; but during swelling, to some extent, the particles become detached from each other or from the substrate in the presence of excess liquid. Another way of immobilization describes the formation of an interparticle crosslinked aggregate, wherein the aggregate is joined to a carrier, which may be comprised of cellulosic fibers or which may be formed by a web. Unfortunately interparticle crosslinking will lead to loss of flexibility and therefore an unpleasant feeling by wearing the absorbent member. Further, inter-particle crosslinked aggregates tend to lose water when stored for a larger period of time and thus the aggregates become stiff or brittle. More flexible absorbent structures are gained by adhesive attachment of superabsorbent particles to fibers. Unfortunately the adhesive attachment negatively influences the absorption profile of the particles and leads to nonuniform swelling.

EP-A-0 700 672 therefore avoids adhesive attachment and proposes the chemical crosslinking for joining superabsorbent particles to the web. The particles of mass median particle size of about 400  $\mu\text{m}$  are applied to a substrate comprising an crosslinking agent.

Unfortunately the basis weights of the superabsorbent particles are very low in order to prevent inter-particle crosslinking and obtaining more flexibility of the composite. By attaching the superabsorbent particles to the substrate in an individual manner, 5 there will be no negative influence to the absorption properties and no detachment of the particles by wetting; moreover, the liquid transmission and flexibility of the substrate is increased. But the highly flexible composite structures comprising a relatively low absorbent capacity are therefore for use in products 10 which are intended to absorb relatively small amounts of fluids, such as panty liners or sanitary napkins or light incontinence pads.

EP-A-0 615 736 proposes containment means which comprise two layers 15 of material which are joined together to form a pocket or compartment, which pocket contains the superabsorbent material. Within the pockets superabsorbent material is incorporated up to 100 w-% without any affixing onto the web; but, in order to prevent gel blocking, the hydrogel-zone is interrupted by the formation 20 of pockets. This proposal requires a lot of effort in the manufacture of such absorbent members and is therefore very expensive.

WO 01/56625 relates to an acquisition layer in an absorbent article 25 which is prepared by forming a sprayable blend comprising (a) one or more superabsorbent-forming monomers, (b) superabsorbent polymer particles, (c) water, and (d) one or more initiators, spraying said sprayable blend onto a fibrous web, and subjecting 30 said fibrous web to conditions under which the superabsorbent forming monomer will polymerize. The absorbent articles are used in disposable hygiene products.

Generally said web is loaded with superabsorbent material up to 60 or 70 w% compared to conventional personal care products. It 35 is desired to provide a matrix wherein superabsorbent polymer is incorporated in amounts convenient to have an optimum in permeability and absorbency and which exhibits wet strength integrity. The absorbent structure has to be of high flexibility and thinness, which is therefore very comfortable in use, even in wearing 40 the personal care products over a longtime period. Therefore it is further desired to provide an absorbent structure having good odor control properties.

It is an object of the present invention to provide an absorbent 45 structure which contains an absorbent layer having a relatively high concentration of high-absorbency material but which absorbent structure is stable and affixed to a matrix and cannot be

detached by mechanical forces. It is desired to provide a matrix wherein superabsorbent polymer is incorporated in amounts convenient to have an optimum in absorbency and storage capacity and which exhibits wet strength integrity. The new absorbent structure has to have a specific absorption capacity especially in regions, where in gush situations the highest amounts of body fluids are applied.

It is further desired to provide an absorbent structure with an unchanged absorption profile after attachment of the superabsorbent material onto the substrate, which is capable of quickly absorbing body fluids applied thereto and afterwards store them. Said superabsorbent polymer exhibits highest absorption rate and absorption capacity without tendency to gel blocking.

15 It is a further object of the invention to provide an absorbent structure of high flexibility and thinness, which is therefore very comfortable in use, even when wearing the personal care products over a long period of time.

20 It is further desired to provide an absorbent structure with an unchanged absorption profile after attachment of the superabsorbent material onto the substrate, which is capable of quickly absorbing body fluids applied thereto and afterwards store them.

25 It is a further object of the invention to provide an absorbent structure of high flexibility and thinness, which is therefore very comfortable in use, even when wearing the personal care products over a long period of time and which exhibits odor control properties.

30 It is further object of the invention to provide an absorbent structure which may contain an absorbent layer having a relatively high concentration of high-absorbency material but which absorbent structure is capable of quickly absorbing body fluids applied thereto, store them, and which exhibits odor control properties. It is a further object of the invention to provide an absorbent structure which retains most of its absorption profile after the application of an odor control material.

40 The objects of the invention are achieved by the use of a layer obtainable by a process comprising

45 (a) forming a sprayable blend comprising one or more superabsorbent forming monomers  
superabsorbent polymer particles

water, and  
one or more initiators,

(b) applying said sprayable blend on a fibrous web; and  
5 (c) subjecting said fibrous web to conditions under which the superabsorbent forming monomer with polymerize,

as a storage layer, preferably by the use of such a layer in hy-  
10 genic products. Such storage layer can be produced according to  
WO 01/56625.

Such a storage layer shows preferably a CRC equal or greater  
9000 g/m<sup>2</sup>, preferable greater 10000 g/m<sup>2</sup>, more preferable greater  
15 12000 g/m<sup>2</sup>.

The objects of the invention are achieved with an absorbent ar-  
ticle comprising at least one double-sided coated fibrous web  
having a storage layer on one side and an acquisition layer on  
20 the other side and/or a combination of at least two adhering  
double-sided coated webs with the proviso that one web has a  
storage layer on both sides and the other web has an acquisition  
layer on both sides wherein said layers are obtained by

25 (a1) forming a blend comprising one or more monomers forming su-  
perabsorbent polymers, superabsorbent polymer particles,  
water, and one or more initiators, said blend having a visco-  
sity of at least 20 mPas (measured at 20°C in a Brookfield  
viscometer, spindle 02, 20 rpm)  
30 (b1) applying said blend onto a fibrous web and  
(c1) carrying out the polymerization of the monomers forming su-  
perabsorbent polymers.

The blend formed in step (a) may further contain a crosslinking  
35 agent and/or a softening agent and/or at least one odor control  
means and/or an agent with skin care effect, e.g. panthotenol,  
aloe vera, a pH in the range of the skin. In the alternative the  
superabsorbent polymer particles of the blend formed in step (a)  
40 may contain at least one odor control means and/or an agent with  
skin care effect, e.g. panthotenol, aloe vera, a pH in the range  
of the skin.

The acquisition layer and storage layer of the absorbent article  
has, for example, a pH of from 2.0 to 7.5 preferably of from 4.0  
45 to 6.5. In an preferred embodiment of the invention the acquisi-  
tion layer and the storage layer have different pH. For example,  
the pH of the acquisition layer may be of from 4.0 to 6.5, pre-

ferably of from 4.2 to 4.5 and the pH of the storage layer may be of from 5.0 to 6.0.

In another preferred embodiment of the invention the superabsorbent particles consist of mixed-bed ion exchange superabsorbent polymers or multi-domain-composites of superabsorbent polymers as described e.g. in WO 99/25393 having preferably an anionic to cationic superabsorbent ratio of from about 5 : 1 to about 1 : 5.

10 The objects of the invention are also achieved with a process for the production of an absorbent article, which comprises

(a1) forming a blend comprising one or more monomers forming superabsorbent polymers, superabsorbent polymer particles, 15 water, and one or more initiators, said blend having a viscosity of at least 20 mPas (measured at 20°C in a Brookfield viscometer, spindle 02, 20 rpm),

(b1) applying said blend onto both sides of at least one fibrous 20 web,

(c1) subjecting said fibrous web to conditions under which the superabsorbent forming monomer will polymerize and

25 (d1) combining by compression at least two coated webs which have different amounts of polymer coated thereon by carrying out steps (b1) and (c1).

The blend formed in step (a1) may further comprises a crosslinking agent and/or odor control means and/or a softening agent 30 and/or skin care agent. If an odor controlling agent is added together with a constituent of the sprayable blend, it is then preferred to use odor control means containing superabsorbent polymer particles. The fibrous web of step (b1) may also contain one 35 or more odor control means and/or skin care agent. It is also possible that the fibrous web as well as the sprayable blend contain one or more odor controlling means.

The addition of one or more softening agents to the blend enables 40 a production of soft and improved skin feel personal care products. Moreover, the whole absorbent web can be rendered to a soft absorbent structure of high flexibility with odor control properties. Nevertheless, said absorbent structure shows good absorbency and good strength. The improved webs retain most of the absorption profile of the absorbent polymers.

The absorbent article may also be prepared by

- (a1) forming a blend comprising one or more monomers forming superabsorbent polymers, superabsorbent polymer particles, 5 water, and one or more initiators, said blend having a viscosity of at least 20 mPas (measured at 20°C in a Brookfield viscometer, spindle 02, 20 rpm),
- (b1) applying said blend onto one side of a fibrous web,
- (c1) carrying out the polymerization of the monomers forming superabsorbent polymers, 10
- (d1) applying the blend formed in step (a1) to the other side of the web in a different amount and
- (e1) carrying out the polymerization of the monomers forming superabsorbent polymer.

15 A higher polymer loading on the web, for example, more than 65 % by weight, with reference to the weight of polymer and web, with up to about 100 % by weight results in storage layers, whereas a polymer loading produced by the process of the invention of less than 65 % by weight, with reference to the polymer and web, results in an acquisition layers. Percentage by weight refers to the total weight of the dried layer with a residual water content of 0 to 5 %, preferably 3 %. In one embodiment of the above process the blend formed in step (a) may further contain a cross-linking agent and/or a softening agent. In another embodiment of the process of the present invention the superabsorbent polymer particles or the blend contain one or more odor control means.

20 The objects of the invention are achieved with an odor control agent containing absorbent article obtained by

- (a2) forming a blend comprising one or more monomers forming superabsorbent polymers, superabsorbent particles, water, one or more initiators, said blend having a viscosity of at least 35 20 mPas (measured at 20°C in a Brookfield viscometer, spindle 02, 20 rpm),
- (b2) applying said blend onto a fibrous web and
- (c2) carrying out the polymerization of the monomers forming superabsorbent polymers in the presence of at least one odor control means, which may be present in either of the blend formed in step (a2) and/or the fibrous web, by subjecting said fibrous web to conditions under which the monomers forming superabsorbent polymers will polymerize or polymerizing 40 45

the blend on the fibrous web and adding thereafter an odor control agent to the absorbent article.

The objects are also achieved with an absorbent article which is 5 obtained by carrying out the above steps (a2) to (c2), when the blend formed in step a2 further comprises a crosslinking agent.

The odor control means can be present in the blend formed in step 10 (a2), i.e. one or more odor control agent is added to the sprayable blend and /or may be contained in the superabsorbent polymer particles or in the softening agent. If an odor controlling agent is added together with a constituent of the sprayable blend, it is then preferred to used odor control means containing superabsorbent polymer particles. The fibrous web of step (b2) may also 15 contain one or more odor control means. It is also possible that the fibrous web as well as the sprayable blend contain one or more odor controlling means.

The obtained absorbent structure has a capability of effectively 20 control odors related to absorbed fluids. The addition of one or more softening agents enables a production of soft and improved skin feel personal care products. Moreover, the whole absorbent web can be rendered to a soft absorbent structure of high flexibility with odor control properties. Nevertheless, said absorbent 25 structure shows good absorbency and good strength. The improved webs retain most of the absorption profile of the absorbent polymers.

The odor control agent containing absorbent article may be prepared 30 by

(a2) forming a blend comprising one or more monomers forming superabsorbent polymers, superabsorbent polymer particles, water, and one or more initiators, said blend having a viscosity of at least 20 mPas (measured at 20°C in a Brookfield 35 viscometer, spindle 02, 20 rpm),

(b2) spraying said sprayable blend onto a fibrous web and

40 (c2) carrying out the polymerization of the monomers forming superabsorbent polymers in the presence of at least one odor control means, which may be present in either of the blend formed in step (a2) and/or the fibrous web, by subjecting 45 said fibrous web to conditions under which the monomers forming superabsorbent polymers will polymerize or polymerizing

the blend on the fibrous web and adding thereafter an odor control agent to the absorbent article.

In one embodiment of the above process the sprayable blend formed in step (a2) further comprises a crosslinking agent. In another embodiment of the process of the present invention the superabsorbent polymer particles contain one or more odor control means.

#### Detailed description of the invention

10

##### Monomers forming superabsorbent polymers

Superabsorbent-polymer-forming monomers, as used herein, are polymerizable compounds which contribute to the absorbency of the polymers formed therefrom. Suitable monomers forming superabsorbent polymers useful in the present invention include monoethylenically unsaturated compounds (or compounds having a polymerizable double bond), having at least one hydrophilic radical, such as carboxyl, carboxylic acid anhydride, carboxylic acid salt, sulfonic acid, sulfonic acid salt, hydroxyl, ether, amide, amino or quaternary ammonium salt groups. Examples of suitable monomers forming superabsorbent polymers are as follows:

1. Carboxyl group-containing monomers: monoethylenically unsaturated mono- or polycarboxylic acids, such as (meth) acrylic acid (meaning acrylic acid or methacrylic acid, similar notations are used hereinafter), maleic acid, fumaric acid; crotonic acid, sorbic acid, itaconic acid, and cinnamic acid.
2. Carboxylic acid anhydride group-containing monomers: monoethylenically unsaturated polycarboxylic acid anhydrides (such as maleic anhydride);
3. Carboxylic acid salt-containing monomers: water-soluble salts (alkali metal salts, ammonium salts, amine salts, etc.) of monoethylenically unsaturated mono- or polycarboxylic acids [such as sodium (meth)acrylate, trimethylamine(meth)acrylate, triethanolamine(meth)acrylate, sodium maleate, methylamine maleate];
4. Sulfonic acid group-containing monomers: aliphatic or aromatic vinyl sulfonic acids (such as vinylsulfonic acid, allyl sulfonic acid, vinyltoluenesulfonic acid, styrene sulfonic acid), (meth)acrylic sulfonic acids [such as sulfopropyl (meth) acrylate, 2-hydroxy-3-(meth)acryloxy propyl sulfonic acid];

5. Sulfonic acid salt group-containing monomers: alkali metal salts, ammonium salts, amine salts of sulfonic acid group-containing monomers as mentioned above.
- 5 6. Hydroxyl group-containing monomers: monoethylenically unsaturated alcohols [such as (meth)allyl alcohol], monoethylenically unsaturated ethers or esters of polyols (alkylene glycols, glycerol, polyoxyalkylene polyols), such as hydroxyethyl (meth)acrylate, hydroxypropyl (meth)acrylate, 10 triethylene glycol (meth)acrylate, poly(oxyethylene oxypropylene) glycol mono (meth)allyl ether (in which hydroxyl groups may be etherified or esterified).
- 15 7. Amide group-containing monomers: vinylformamide, (meth)acrylamide, N-alkyl (meth)acrylamides (such as N-methylacrylamide, N-hexylacrylamide), N,N-dialkyl (meth)acrylamides (such as N,N-dimethylacrylamide, N,N-di-n-propylacrylamide), N-hydroxyalkyl (meth)acrylamides [such as N-methylol (meth)acrylamide, N-hydroxyethyl (meth)acrylamide], N,N-dihydroxyalkyl (meth)acrylamides [such as N,N-dihydroxyethyl (meth)acrylamide], vinyl lactams (such as N-vinylpyrrolidone);
- 20 8. Amino group-containing monomers: amino group-containing esters (e.g. dialkylaminoalkyl esters, dihydroxyalkylaminoalkyl esters, morpholinoalkyl esters, etc.) of monoethylenically unsaturated mono- or dicarboxylic acid [such as dimethylaminoethyl (meth)acrylate, diethylaminoethyl (meth)acrylate, morpholinoethyl (meth)acrylate, dimethylaminoethyl fumarate], heterocyclic vinyl compounds [such as 30 vinyl pyridines (e.g. 2-vinyl pyridine, 4-vinyl pyridine, N-vinyl pyridine), N-vinyl imidazol]; and
- 35 9. Quaternary ammonium salt group-containing monomers: N,N,N-trialkyl-N-(meth)acryloyloxyalkylammonium salts [such as N,N,N-trimethyl-N-(meth)acryloyloxyethylammonium chloride, N,N,N-triethyl-N-(meth)acryloyloxyethylammonium chloride, 2-hydroxy-3-(meth)acryloyloxypropyl trimethyl ammonium chloride], and monomers as mentioned in British patent specification No. 1,034,296.
- 40 45 Suitable monomers which become water-soluble by hydrolysis, for use in this invention instead of or in conjunction with the water-soluble monomers, include monoethylenically unsaturated compounds having at least one hydrolyzable group, such as esters, amide and nitrile groups. Such monomers having an ester group include for example, lower alkyl (C<sub>1</sub>-C<sub>4</sub>) esters of monoethylenically unsaturated carboxylic acids, such as methyl (meth)acrylate,

ethyl (meth)acrylate and 2-ethylhexyl (meth) acrylate; and esters of monoethylenically unsaturated alcohols [vinyl esters, (meth)-allyl ester, etc.], such as vinyl acetate and (meth) allyl acetate. Suitable nitrile group-containing monomers include 5 (meth) acrylonitrile.

Among these monomers having a polymerizable double bond which are water-soluble or become water-soluble by hydrolysis, water-soluble monomers which do not need hydrolysis after polymerization 10 are preferred from the viewpoint of providing an easy process for producing water-absorbing resins. Further, from the viewpoint of providing water-absorbing resins having higher water-absorbance, the preferred water-soluble monomers are carboxyl group-containing monomers such as (meth)-acrylic acid and maleic acid anhydride; carboxyl acid salt group-containing monomers such as sodium (meth)acrylate, trimethylamine (meth)acrylate and triethanolamine (meth)acrylate, and quaternary ammonium salt group-containing monomers such as N,N,N-trimethyl-N-(meth)acryloyloxyethylammonium chloride. Most preferred superabsorbent forming monomers in the present invention include, for example, acrylic acid, methacrylic acid, maleic acid, fumaric acid, crotonic acid, sorbic acid, itaconic acid, cinnamic acid, vinyl sulfonic acid, allyl sulfonic acid, vinyl toluene sulfonic acid, styrene sulfonic acid, sulfo(meth)acrylate, sulfopropyl(meth)acrylate, 25 2-acrylamid-2-methylpropane sulfonic acid, 2-hydroxyethyl(meth)acryloylphosphate, phenyl-2-acryloyloxyethylphosphate, the sodium, potassium and ammonium salts thereof, maleic anhydride and combinations thereof. It is also preferred that the superabsorbent forming monomer in the sprayable blend is at least 30 partially neutralized, preferably neutralized to a level of from 1 to 100 mole percent, more preferably from 10 to 80 mole percent, and most preferably from 15 to 75 mole percent. Most preferably, the superabsorbent forming monomer is neutralized acrylic acid.

35 The monomers forming superabsorbent polymers are present in the blends formed in step (a) at a level of from about 15 to 50 percent by weight, preferably from 17 to 40 percent by weight, most preferably from 20 to 35 percent by weight. If the level of the 40 monomers forming superabsorbent polymers in said blend is too low, the resulting web is likely to have poor performance characteristics. Within the preferred ranges, the conversion of superabsorbent-polymer-forming monomers into polymer, when subjected to polymerization conditions, is generally much better. Also, increasing the relative amount of the monomers forming superabsorbent polymers in the blend formed in step (a) generally reduces the amount of water in the blend. This is beneficial since it re-

quires time, energy and expense to remove additional water from the final web. When partially neutralized acrylic acid is used as monomer forming superabsorbent polymers, e.g. 75% neutralized, it is preferred to use the partially neutralized acrylic acid in the 5 blend in a concentration of no more than 38% by weight.

#### Superabsorbent polymer particles

Superabsorbent polymer particles are lightly crosslinked polymers 10 capable of absorbing several times their own weight in water and/or saline. Superabsorbent polymer particles can be made by any conventional process for preparing superabsorbent polymers and are well known to those skilled in the art. Suitable process for preparing superabsorbent polymer particles include the processes described in U.S. Pat. Nos. 4,076,663; 4,286,082; 15 4,654,039 and 5,145,906 which describe the solution polymerization method and U.S. Pat. Nos. 4,340,706; 4,497,930; 4,666,975; 4,507,438 and 4,683,274 which describe the inverse suspension polymerization method, the disclosures of which are hereby incorporated by reference. Preferred superabsorbent polymer particles 20 have an average particle size which is small enough so that the particles do not clog the spray equipment, preferably below about 150  $\mu\text{m}$ , more preferably below about 100  $\mu\text{m}$ . Such particle size can be obtained directly as a result of the polymerization process, 25 or superabsorbent polymers can be sieved, ground, pulverized, attrited or a combination thereof to achieve superabsorbent polymer particles having the desired average particle size. The mean particle size diameter of the superabsorber polymer particles is, for example, in the range from 10 to 130  $\mu\text{m}$  preferably 15 to 30 100  $\mu\text{m}$  and most preferably 40 to 90  $\mu\text{m}$ .

The superabsorbent polymer particles are present in the blends formed in step (a1) or (a2) at a level of from about 1 to 20 percent by weight, preferably from 2 to 15 percent by weight, most 35 preferably from 5 to 10 percent by weight. It has been observed that if the level of superabsorbent polymer particles is too high, premature polymerization can occur in the blend even in the absence of any initiators.

40 Superabsorbent polymer particles useful in the present invention are prepared from one or more monoethylenically, unsaturated, water- soluble carboxyl or carboxylic acid anhydride containing monomers and the alkali metal and ammonium salts thereof wherein said monomers comprise 50 to 99.9 mole percent of said polymer. 45 Exemplary monomers include acrylic acid, methacrylic acid, maleic acid, fumaric acid, maleic anhydride and the sodium, potassium

and ammonium salts thereof. The preferred monomer is acrylic acid.

Monoethylenically, unsaturated monomers are polymerized in the presence of an internal crosslinking compound to provide a lightly crosslinked base polymer wherein the crosslinking is substantially uniform throughout the polymer particles. These internal crosslinkers are well known in the art. Suitable crosslinkers are those compounds having two or more groups capable of reacting with the monoethylenically unsaturated monomers and which are at least partially water soluble or water dispersible, or at least partially soluble or dispersible in an aqueous monomer mixture. The internal crosslinking compound may be selected from a polyunsaturated monomer such as divinylbenzene, a compound having at least two functional groups which are reactive with the monoethylenically unsaturated monomer such as ethylenediamine, a compound having at least one unsaturated bond and at least one reactive functional group such as glycidyl acrylate.

Exemplary internal crosslinkers are: tetraallyloxyethane, N,N'-methylene bisacrylamide, N,N'-methylene bismethacrylamide, triallylamine, trimethylol propane triacrylate, glycerol propoxy triacrylate, divinylbenzene, N-methylol acrylamide, N-methylol methacrylamide, glycidyl methacrylate, polyethylene polyamines, ethyl diamine, ethyl glycol, glycerine and the like. Preferred internal crosslinking monomers are those containing at least two allyl groups, most preferably three or four allyl groups. Preferred internal crosslinkers are tetraallyloxyethane and triallyl ether of pentaerythritol. The amount of internal crosslinker employed in the invention will depend on the internal crosslinker and the polymerization method. Generally the amount of internal crosslinker will vary from about 0.005 to about 1.0 mole percent based on moles of ethylenically unsaturated monomer.

Optional components used in the preparation of the superabsorbent polymer particles are water soluble hydroxy group containing polymers, such as polysaccharides and vinyl or acrylic polymers. Examples of water soluble polysaccharides are starches, water soluble celluloses and polygalactomannans. Suitable starches include the natural starches, such as sweet potato starch, potato starch, wheat starch, corn starch, rice starch, tapioca starch and the like. Processed or modified starches, such as dialdehyde starch, alkyl-etherified starch, allyl-etherified starch, oxyalkylated starch, aminoethyl-etherified starch, and cyanomethyl-etherified starch are also suitable. Polyvinyl alcohol and polyvinyl alcohol copolymers are also suitable.

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The water-soluble celluloses useful in this invention are those obtained from such sources as wood, stems, bast, seed fluffs and the like which are then derivatized to form hydroxyalkyl cellulose, carboxymethyl cellulose, methyl cellulose and the like.

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Suitable polygalactomannans are guar gum and locust bean gums as well as their hydroxyalkyl, carboxyalkyl, and aminoalkyl derivatives. Water soluble vinyl and acrylic polymers include polyvinyl alcohol and poly(hydroxyethyl acrylate). The preferred polysac-  
10 charide for use in this invention is natural starch, such as wheat starch, corn starch and alpha starches. These optional pre-formed hydroxy containing polymers may be used in an amount from about 1 to 15 percent, preferably about 1 to 10 percent, most preferably about 1 to 5 percent.

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The superabsorbent polymer particles useful in the present invention may be prepared by well known polymerization methods. The polymerization reaction is conducted in the presence of, for example, redox initiators and thermal initiators. The redox initia-  
20 tors can be used as the primary initiator with the thermal poly-  
merization initiators being used if desired to reduce the free monomer content of the final polymerization product below 0.1 percent by weight. Optionally, thermal initiators or redox initiators may be used as the sole initiator system. Examples of dif-  
25 ferent initiator systems are found in U.S. Pat. No. 4,497,930 which discloses a two component initiator system comprising a persulfate and a hydroperoxide and U.S. Pat. No. 5,145,906 which discloses a three component initiator system; i.e. redox system plus thermal initiator.

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Any of the well known water soluble reducing agents and oxidizing agent can be used in this invention as the redox initiator. Examples of reducing agents include such compounds as ascorbic acid, alkali metal sulfites, alkali metal bisulfites, ammonium sulfite, ammonium bisulfite, alkali metal hydrogen sulfite, ammonium hy-  
35 drogen sulfite, ferrous metal salts, e.g. ferrous sulfates, sug-  
ars, aldehydes, primary and secondary alcohols, and the like.

Oxidizing agents include such compounds as hydrogen peroxide, al-  
40 kali metal persulfates, ammonium persulfates, alkylhydroperox-  
ides, peresters, diacryl peroxides, silver salts, and the like. A particularly preferred redox initiator pair is ascorbic acid and hydrogen peroxide. The reducing agent is used, for example, in an amount of about  $2 \times 10^{-5}$  to about  $2 \times 10^{-2}$  mole percent based on  
45 moles of acrylic acid.

In order to ensure complete polymerization of the unsaturated monomer and the crosslinking monomer, a thermal initiator may be included in the polymerization process. Useful thermal initiators are the "azo" initiators, i.e. compounds which contain the -N=N- structure. Any of the azo compounds which have solubility in water or in a monomer-water mixture and which have an 10 hour half life at 30 °C or above can be used. Examples of useful azo initiators are 2,2'-azobis(amidino) propane dihydrochloride, 4,4'-azo-bis(cyanovaleric acid), 4,4'-butylazo-cyanovaleric acid, 5 2,2'-azobis(isobutyronitrile), 2,2'-azobis(2-(2-imidazole-2-yl))propane dihydrochloride, and the like. Other thermal initiators include the persulfates and hydroperoxides when used in the absence of a reducing agent, e.g. sodium, potassium and ammonium persulfates, t-butylhydroperoxide and the like. A preferred azo initiator for use in this invention is 2,2'-azo-bis(amidino) propane dihydrochloride. The thermal initiators are used in the amount of 0 to about 1 mole percent based on the weight of unsaturated monomer.

20 The superabsorbent polymer may be prepared by the solution or the inverse suspension polymerization method or any suitable bulk polymerization method. The solution polymerization and inverse polymerization methods are well known in the art; see for example U.S. Pat. Nos. 4,076,663; 4,286,082; 4,654,039 and 5,145,906 25 which describe the solution polymerization method and U.S. Pat. Nos. 4,340,706; 4,497,930; 4,666,975; 4,507,438 and 4,683,274 which describe the inverse suspension polymerization method. The teachings of these patents are hereby incorporated by reference.

30 In the solution polymerization method, the water soluble monomer is polymerized at a monomer concentration from about 5 to about 30 percent in aqueous solution at a temperature from about 5 °C to about 150 °C depending upon the polymerization initiator system. A detailed description of the solution polymerization method is 35 given in U.S. Pat. No. 5,145,906; The teachings of which are hereby incorporated by reference.

In the inverse suspension polymerization process, the unsaturated monomer in an aqueous solution (about 35 to 60 percent monomer to 40 65 to 40 percent water) is dispersed in an alicyclic or aliphatic hydrocarbon suspension medium in the presence of a dispersing agent, such as a surfactant or protective colloid such as polyvinyl alcohol. A surfactant having a HLB value of 8 to 12 such as a sorbitan fatty acid ester may be employed as the dispersing 45 agent. The inverse suspension polymerization method is described

in detail in U.S. Pat. No. 4,340,706; The teachings of which are hereby incorporated by reference.

The carboxylic acid groups or the unsaturated monomer used in the 5 polymerization may be neutralized prior to or subsequent to the polymerization. Suitable neutralizing agents include an alkali such as sodium hydroxide, ammonium hydroxide, potassium hydroxide or the like, and the appropriate degree of neutralization is 50 - 98 mole percent; preferably 60 - 75 mole percent. The degree of 10 neutralization is preferably at least 50 mole percent. Low neutralization levels (less than 50 mole percent) tend to result in superabsorbent polymers having lower absorbency properties.

The polymer is prepared by either the solution or inverse suspension 15 polymerization method dried and screened to provide a superabsorbent particle with an appropriate particle size distribution and particle shape. Generally the superabsorbent particle size distribution (mean particle diameter) should be between 10 and 300  $\mu\text{m}$ , preferably between 45 and 150  $\mu\text{m}$  and most preferably 50 to 20 90  $\mu\text{m}$ . Large particles over 300  $\mu\text{m}$  are undesired since they may tend to clog the nozzle of spray equipment used to spray the sprayable blend. Also, large particles may cause the absorbent article to have an abrasive feel.

25 Although many of the conventional teachings in this area prefer to avoid the use of small particle size superabsorbent polymer particles, we have found that these particles work quite well in the present invention. Superabsorbent polymer particles having particle sizes in the range of from 10 to 300  $\mu\text{m}$  tend to work 30 well with the spray equipment, produce sprayable blends having manageable viscosities, and result in absorbent articles having discrete superabsorbent particles attached to the fibrous web.

The superabsorbent polymer particles described above may option- 35 ally be further treated with a crosslinker solution containing from about 0.5 to about 3.5 weight percent water, from about 1.0 to 2.5 weight percent of a water miscible solvent selected from a C3 to C6 diol and a crosslinker having at least two functional groups that can react with the carboxyl, carboxylate or other 40 reactive groups in the superabsorbent polymer chain to crosslink the polymer chains on or in the vicinity of the surface of superabsorbent polymer particles. The term "diol" is intended to mean a dihydroxy aliphatic compound which may be a linear or branched compound, i.e. a glycol. The term "surface crosslinking" is used 45 to describe this process of crosslinking the polymer chains on or in the vicinity of the particle's surface. The terms, "surface crosslinker" and "surface crosslinker solution" are likewise used

to describe the crosslinking compound and the solution used to effect this surface crosslinking process. The crosslinking compound is used, for example, in an amount of from about 0.01 to about 3 weight percent, preferably 0.1 to 1.5 weight percent and 5 most preferably 0.25 to 1 weight percent, based upon the weight of the superabsorbent polymer. The surface crosslinker may be selected from diglycidyl ethers, haloepoxy, isocyanate, aziridinyl, azetidinium group containing compounds, oxazolidone, polyamine, polyamine-polyamide, polyamine-epichlorohydrin adducts and amine- 10 polymer-epichlorohydrin adducts and the like. Preferred crosslinkers are diglycidyl ether compounds having a molecular weight of at least 200 and the polymeric epichlorohydrin adducts having a molecular weight average in excess of 2000.

15 Exemplary surface crosslinkers are poly(ethylene glycol) diglycidyl ethers, poly(propylene glycol) diglycidyl ethers, epichlorohydrin substituted compounds, methyl-epichlorohydrin substituted compounds, hexamethylene diisocyanate, triethylene triamine, polyethylene amine, 2-oxazolidone, 2,2-bishydroxymethylbutanol- 20 tris[3-(1-azindinyl)propionate], polyamine epichlorohydrin adducts, polyethylene- polyamine-epichlorohydrin adducts, and the like.

When used, the preferred surface crosslinkers are the higher molecular weight diglycidyl ether compounds, polyamide (polyamide-polyamine) epichlorohydrin adducts, polyamine-epichlorohydrin adducts and amine polymer epichlorohydrin adducts. Polyamide-epichlorohydrin adducts are prepared by reacting epichlorohydrin with the polycondensation product of a polyalkylene polyamine 30 with a polycarboxylic acid such as diethylene triamine with a dibasic acid such as adipic acid. Polyamine epichlorohydrin adducts are made by condensing a polyalkylene polyamine directly with epichlorohydrin. These adducts include polyalkylene polyamines which are linked together with dihalides to form higher poly- 35 amines before reacting them with epichlorohydrin. Amine polymer epichlorohydrin adducts include resins in which the monomeric amine is polymerized to a polyamine precursor which is then alkylated and reacted with epichlorohydrin. They include amines substituted polymers of vinyl, allyl, acrylate or epoxy monomers. 40 The epichlorohydrin adducts whether the polymer is polyamide, a polyamine or an amine polymer react with the epichlorohydrin by different routes. If the amino group in the polymer chain is a primary amine, two epichlorohydrin molecules reacted with the nitrogen and form a disubstituted chlorohydroxypropyl substituted 45 amine group. Secondary amine groups react with epichlorohydrin to form a tertiary aminochlorohydrin group which gives a reactive 3-hydroxyazetidinium salt moiety. This is a preferred reactive

group. Tertiary amine groups react with epichlorohydrin to form a glycidyl; (2,3-epoxypropyl) ammonium salt. Preferably the reactive group is an azetidinium moiety. However, these adducts may contain a mixture of chlorohydroxypropyl, epoxy and azetidinium groups. Preferably the epichlorohydrin adducts have a molecular weight of at least 2.000; preferably 300.000 to 500.000 and wherein at least 50 mole percent of the reactive groups in the adduct are the azetidinium groups. A preferred polymer is one in which about 90% of the substitution is in the form of an azetidinium group and about 10% as an epoxide group. Exemplary products are Reten.RTM.204LS and Kymene.RTM.736 epichlorohydrin adducts; available from Hercules Inc., Wilmington, Del. These products are sold as an aqueous solution of the reactive epichlorohydrin adduct. The Reten.RTM.204LS product as a 38% aqueous solution.

15 The surface crosslinker solution should have a surface tension not greater than about 55 dynes per cm; preferably not greater than about 50 dynes per cm; e.g. about 40 to 50 dynes per cm. In the event the surface tension of the crosslinker solution is higher than about 55 dynes per cm; the surface crosslinked polymer has inferior absorbency as evidenced by a low 0.6 psi AUL value. While not being bound to any theory, it is believed that when the surface tension of the crosslinker solution is higher than about 55 dynes per cm, the solution is not uniformly distributed on the surface of the polymer particles and a lower absorbency value results. Optionally, a surfactant may be used to reduce the surface tension of the crosslinker solution.

The desired surface tension is achieved by adding the C<sub>3</sub> to C<sub>6</sub> dihydroxy compound to water component of the crosslinker solution to achieve a surface tension below about 55 dynes per cm range. The amount of each solvent is determined by simple experimentation. Generally the crosslinker has a negligible effect on the surface tension of the crosslinker solution. The diols useful in the invention are propylene glycol, butylene glycol, pentanediol and hexanediol. Ethylene glycol was found to be an undesired solvent because it tends to swell the superabsorbent polymer particles and their surface becomes tacky which results in undesired particle agglomeration. In addition, ethylene glycol is undesirable because of its toxicity and biodegradability properties. The C<sub>3</sub> to C<sub>6</sub> diol is used in an amount of from about 1 percent by weight to about 2.5 percent by weight based upon the weight of superabsorbent polymer; preferably about 1 to about 2 percent by weight. The water component of the crosslinker solution comprises about 0.5 to 3.5 percent by water based upon the weight of the polymer, preferably about 1.5 to 2.0 percent.

The total amount of crosslinker solution used depends upon the type of equipment and the method used to coat the base polymer with the surface crosslinking solution. Generally the amount of crosslinker solution should be about 1.5% minimum based of the weight of the polymer. The crosslinker solution is applied to the base polymer particles in a manner such that the solution is uniformly distributed on the surface of the base polymer particle. Any of the known methods for dispersing a liquid can be used; preferably by dispersing the crosslinker solution into fine droplets; e.g. by use of a pressurized nozzle or a rotating disc. Uniform crosslinker dispersion on the base polymer can be achieved in a high intensity mechanical mixer or a fluidized mixture which suspends the base polymer in a turbulent gas stream. Methods for the dispersion of a liquid onto the superabsorbent base polymer's surface are known in the art; see for example U.S. Patent 4,734,478; the teachings of which are hereby incorporated by reference; in particular column 6, line 45 to column 7, line 35.

Exemplary commercially available equipment for conducting the crosslinker solution dispersion step of the invention are high speed variable intensity paddle mixers such as the "Turbulizer" mixer of the Bepex Corporation, Rolling Meadows, Ill. Or the high speed variable intensity vertical mixer sold by Bepex under the tradename, "Turboflex". These machines are generally operated in a continuous manner using a short residence time in the order of 2 seconds to 2 minutes, typically 2-30 seconds. Dispersion may be effected batchwise in a high intensity mixer such as a Henschel mixer or in liquid-solid V-blender equipped with a liquid dispersion device. In any event, whether a batchwise or continuous dispersion method is used, simple experimentation can be conducted to determine the best process conditions for the particular machine employed in the process. Preferably, the surface crosslinker is coated onto the polymer particles under high intensity mixing conditions.

After effecting dispersion of the surface crosslinker on the base polymer particle the crosslinking reaction is effected and the polymer particle dried. The crosslinking reaction may be effected at a temperature from about 70°C to about 220°C, preferably to about 180°C.

In a preferred embodiment of the invention a mixture of cationic and anionic hydrophilic absorbent polymers ("bipolar absorbent polymers") are used. Such polymers are generally obtained by mixing cation-exchange absorbent polymers with anion-exchange absorbent polymers. Cation-exchange polymers are for example, ethylene

maleic anhydride copolymer, crosslinked polyvinyl sulphonic acid, crosslinked polyacrylic acid, or the graft copolymers of said monomers with starch or cellulose based polymer. The functional groups are selected from sulphonic, sulphate, phosphonate or carboxyl. Preferred cation-exchange based polymers are crosslinked polyacrylates or crosslinked isobutylene maleic anhydride copolymer. Anion-exchange polymers contain primary, secondary or tertiary amine or quaternary ammonium groups, for example polyvinylamine, polyallylamine, polyethyleneimine and polydiallyldimethylammoniumhydroxide. The superabsorbent particles have a anionic to cationic superabsorbent ratio from about 5 : 1 to about 1 : 5. Such products are, for example, disclosed in U.S. Patent 6,222,091 and WO-A-98/37149 (both incorporated by reference).

### 15 Odor control means

In one embodiment of the invention superabsorbent polymer particles contain odor control means. Each of the odor control materials known in the art can be used. Further it is possible to use mixtures thereof. Odor control materials known in the art include zeolites, silica, carbon, chelants, antimicrobial agents, perfuming ingredients, masking agents, and mixtures thereof, for example:

25 - Inorganic materials for odor control and absorption at the same time ( e. g. zeolites, activated carbon, bentonite, silica (AEROSIL® or CAB-O-SIL®), aerosile, clays) have relatively high surface areas. They are added in form of powder or granulate to the surface of superabsorbent polymer particles.

30 - Chelants prevent malodors by complexing organic molecules or metal ions present in body fluids. Preferably ethylene diaminetetraacetic acid, cyclodextrin (alpha-cyclodextrin, beta-cyclodextrin, gamma-cyclodextrin and / or their derivatives and / or mixtures thereof) are used. Most preferably mixtures of cyclodextrins are used. Generally, cyclodextrins may have particle sizes between 12 to 150  $\mu\text{m}$  or are added in particle sizes of lower than 12  $\mu\text{m}$  to have highest surface areas for odor control. Further chelants are aminopolycarbonic acids and their salts, ethylenediamine pentamethylene phosphonic acid, ethylenediamine tetramethylene phosphonic acid, amino-phosphate, polyfunctional aromates, N,N-disuccinic acid.

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45 - Perfuming ingredients. Nonlimiting examples of preferred volatile perfume ingredients are Allo-ocimene, allyl caproate, allyl cyclohexaneacetate, allyl cyclohexanopropionate, allyl

heptanoate, amyl acetate, amyl propionate, anethol, anixic aldehyde, anisole, benzaldehyde, benzyl acetate, benzyl acetone, benzyl alcohole, benzyl butyrate, benzyl formate, benzyl isovalerate, benzyl propionate,  $\beta$ - $\gamma$ -hexenol, butyl benzoate, butyl caproate, 4-tert-butylcyclohexyl formate, camphene, camphor gum, carvacrol, laevo-carveol, d-carvone, laevo-carvone, cinnamyl formate, cis-jasmone, cis-3-hexenyl acetate, cis-3-hexenyl butyrate, cis-3-hexenyl caproate, cis-3-hexenyl tiglate, cis-3-hexenyl valerate, citral, citronellol, citronellyl acetate, citronellyl formate, citronellyl isobutyrate, citronellyl nitrile, citronellyl oxyacetaldehyde, citronellyl propionate, cuminic alcohol, cuminic aldehyde, Cyclal C, cyclohexyl ethylacetate,  $\alpha$ -damascone, 2-decenale, decyl aldehyde, dihydro myrcenol, dihydro myrcenyl acetate, dimethyl benzyl carbinol, dimethyl benzyl carbinal acetate, dimethyl benzyl carbinal propionate, dimethyl phenylethyl carbinal acetate, 3,7-dimethyl octanal, dimethyl octanol, diphenyl oxide, ethyl acetate, ethyl aceto acetate, ethyl amyl ketone, ethyl benzoate, ethyl butyrate, ethyl hexyl ketone, ethyl phenyl acetate, eucalyptol, fenchyl acetate, fenchyl alcohole,  $\alpha$ -methyl-ionon,  $\alpha$ -nonalactone, geraniol, geranyl acetate, geranyl acetoacetate, geranyl butyrate, geranyl formate, geranyl isobutyrate, geranyl nitrile, geranyl propionate, heptyl acetate, heptyl isobutyrate, heptyl propionate, hexenol, hexenyl acetate, hexenyl isobutyrate, hexyl acetate, hexyl formate, hexyl isobutyrate, hexyl isovalerate, hexyl neopentanoate, hexyl tiglate, hydratropic alcohol, hydroxy citronellal,  $\alpha$ -ionone,  $\beta$ -ionone,  $\gamma$ -ionone,  $\alpha$ -irone, isoamyl alcohol, isobornyl acetate, isobornyl propionate, isobutyl benzoate, isobutyl caproate, isononyl acetate, isononyl alcohol, isomenthol, isomenthone, isononyl acetate, isopulegol, isopulegyl acetate, isoquinoline, lauric aldehyde, lavandulyl acetate, ligustral, d-limonene, linalool, linalool oxide, linalyl acetate, linalyl butyrate, linalyl isobutyrate, linalyl formate, linalyl propionate, menthone, menthyl acetate, methyl acetophenone, methyl amyl ketone, methyl anthranilate, methyl benzoate, methyl benzyl acetate, methyl chavicol, methyl eugenol, methyl heptenone, methyl heptine carbonate, methyl heptyl ketone, methyl hexyl ketone, methyl nonyl acetaldehyde,  $\alpha$ -iso" $\gamma$ "methyl ionone, methyl octyl acetaldehyde, methyl octylketone, methyl phenyl carbinal acetate, methyl salicylate, myrcene, myrcenyl acetate, neral, nerol, neryl acetate, nonalactone, nonyl butyrate, nonyl alcohol, nonyl acetate, nonyl aldehyde, octalactone, octyl acetate, octyl alcohol, octyl aldehyde, orange terpenes, p-cresol, p-cresyl methyl ether, p-cymene, p-isopropyl-p-methyl acetophenone, phenethyl anthranilate, phenoxy

ethanol, phenyl acetaldehyde, phenyl ethyl acetate, phenyl ethyl alcohol, phenyl ethyl dimethyl carbinol,  $\alpha$ -pinene,  $\beta$ -pinene, prenyl acetate, propyl butyrate, pulegone, rose oxide, safrole,  $\alpha$ -terpinene,  $\gamma$ -terpinene, 4-terpineol, terpineol, terpinolene, terpinyl acetate, terpinyl propionate, tetrahydro linalool, tetrahydro myrcenol, thymol, tricyclo decenyl acetate, tricyclo decenyl propionate,  $\delta$ -undecalactone,  $\gamma$ -undecalactone, undecanal, undecenal, undecyl alcohol, Veratrol, Verdox, Vertenex, viridine.

10 - Antimicrobial agent may be any chemical capable of preventing the growth of or killing microorganisms, and is capable of preferably binding to the surface superabsorbent material. Preferred antimicrobials are those that can prevent the growth of or kill microorganisms typically found in body fluids, more preferably those body fluids typically collected by a disposable absorbent article. Preferred antimicrobials include, but are not limited to, quarternary ammonium, phenolic, amide, acid, and nitro compounds, and mixtures thereof; more preferably quarternary ammonium, acid and phenolic; more preferably still quarternary ammonium compounds. Preferred quarternary ammonium compounds include, but are not limited to, 2-(3-anilinovinyl)3,4-dimethyl-oxazolium iodide, alkylisoquinolium bromide, benzalkonium chloride, benzethonium chloride, cetylpyridinium chloride, chlorhexidine gluconate, chlorhexidine hydrochloride, lauryl trimethyl ammonium compounds, methylbenzethonium chloride, stearyltrimethylammonium chloride, and mixtures thereof; more preferably benzalkonium chloride, chlorhexidine gluconate, and 3-(trimethoxysilyl)propyldimethyloctadecyl ammonium chloride.

15 - Preferred phenolic compounds include, but are not limited to, benzyl alcohol, p-chlorophenol, chlorocresol, chloroxylenol, cresol, o-cymene-5-ol (BIOSOL), hexachlorophene, hinokitiol, isopropylmethylphenol, parabens (having methyl, ethyl, propyl, butyl, isobutyl, isopropyl, and/or sodium methyl substituents), phenethyl alcohol, phenol, phenoxyethanol, o-phynylphenol, resorcin, resorcin monoacetate, sodium parabens, sodium phenolsulfonate, thioxolone, 2,4,4'-trichloro-2'-hydroxydiphenyl ether, zinc phenolsulfonate, di-tert.-butyl phenole, hydrochinone, and mixtures thereof, more preferably sodium parabens.

20 - Preferred amides include, but are not limited to, diazolidinyl urea, 2,4-imidazolidinedione (HYDATOIN), 3,4,4'-trichlorocarbanilide, 3-trifluoromethyl-4,4'-dichlorocarbanilide,

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undecylenic acid monoethanolamide, and mixtures thereof, more preferably still 2,4-imidazolidinedione.

Preferred acids include, but are not limited to, ascorbic acid, benzoate, benzoic acid, citric acid, dehydroacetic acid, potassium sorbate, salicylic acid derivatives (acetyl salicylic acid, salicylic acid aldehyde), sodium citrate, sodium dehydroacetate, sodium salicylate, sodium salicylic acid, sorbic acid, undecylenic acid, zinc undecylenate, and mixtures thereof, more preferably benzoic acid, citric acid, salicylic acid and sorbic acid, more preferably still citric acid and sorbic acid.

Preferred nitro compounds include, but are not limited to, 2-bromo-2-nitro-2,3-propanediol (BRONOPOL), and methyldibromo glutaronitrile and propylene glycol (MERGUARD), and mixtures thereof.

Further preferred antimicrobial agents are 2,5-dimethoxy tetrahydrofuran, 2,5-diethoxy tetrahydrofuran, 2,5-dimethoxy-2,5-dihydrofuran, 2,5-diethoxy-2,5-dihydrofuran, succine dialdehyde, glutar dialdehyde, glyoxal, glyoxylic acid, hexahydrotriazine, tetrahydro-3,5-dimethyl-2H-1,3,5-thiadiazine-2-thione (Dazomet), 2,4-dichlorobenzyl alcohol, benzalkonium chloride, chlorhexidine gluconate, triclosan.

Masking agents: coazervate perfume encapsulation is a commonly known encapsulation method. Small amounts of one or more perfume ingredients listed above are enclosed in a solid wall material. Water-soluble cellular marix microcapsules are described in detail in U.S. Patent 5,429,628 which is incorporated herein by reference. Water-soluble cellular marix microcapsules are used for time-delay release of the perfume ingredient.

35 - Addition of compounds of transition metals such as Cu, Ag, Zn.  
- Addition of enzymes (Urease-inhibitors)

40 Further compounds for odor control are low pH-derivatives, peroxides, hydrogencarbonates, extracts of vegetables, etherical oils, boron derivatives, poly-alpha-aminoacids, imides, polyimides, pvp-jode, chitosane, polyglycoside, cyclophanes.

Odor control means in particle and / or powder form and / or aqueous solution comprising

- optionally a solubilized uncomplexed cyclodextrin to absorb
- 5 malodors
- optionally a antimicrobial agent to reduce growth of bacteria
- optionally a perfume ingredient, optionally with masking
- 10 agents: Small amounts of one or more perfume ingredients listed above are enclosed in water-soluble cellular marix microcapsules for timed release of the perfume ingredient.
- optionally an enzyme to improve odor control benefit.

15 Preferably used odor control means are selected from the group consisting of zeolites, bentonite, silica, cyclodextrins, amino-polycarboxylic acids, perfumes, antimicrobial agents, and enzymes.

20 Another group of odor control means are polymers containing acidic groups and/or anhydride groups. These polymers differ from superabsorbent polymers in that they are not crosslinked. Their pH is below 6.5, preferably below 5.5, for example in the range 25 of from pH 4.2 to pH 5.0.

Suitable monomers for the preparation of the uncrosslinked, acidic and/or anhydride groups containing polymers are those listed above under Superabsorbent-polymer-forming monomer. Preferred are 30 polymerizable, unsaturated, acidic groups containing monomers. Examples of olefinically unsaturated carboxylic acid and carboxylic acid anhydride monomers are monoethylenically unsaturated C<sub>3</sub>-bis C<sub>25</sub>- carboxylic acids or anhydrides as acrylic acid, methacrylic acid, ethacrylic acid, crotonic acid,  $\alpha$ -acryloxypropionic acid, sorbic acid,  $\alpha$ -stearylacrylic acid, maleic acid, maleic acid anhydride, itaconic acid, citraconic acid, mesaconic acid, glutaconic acid, aconitic acid, fumaric acid, itaconic acid anhydride and maleic acid anhydride.

40 Examples for olefinically unsaturated sulfonic acid and phosphonic acid monomers include aliphatic or aromatic vinyl sulfonic acids such as vinylsulfonic acid, allyl sulfonic acid, vinyl toluene sulfonic acid and styrene sulfonic acid; acrylic and methacrylic sulfonic acid such as sulfoethyl acrylate, sulfoethyl methacrylate, sulfopropyl acrylate, sulfopropyl methacrylate, styrene sulfonic acid, 2-hydroxy-3-acryloxypropyl sulfonic acid, 2-hydroxy-3-methacryloxypropyl sulfonic acid, 2-acrylamide-2-me-

thylpropane sulfonic acid; vinylphosphonic acid, allylphosphonic acid and mixtures thereof.

Preferred monomers are acrylic acid, methacrylic acid, vinylsulfonic acid, 2-acrylamide-2-methylpropane sulfonic acid and mixtures thereof, such as mixtures of acrylic acid and methacrylic acid, mixtures of acrylic acid and 2-acrylamide-2-methylpropane sulfonic acid or mixtures of acrylic acid and vinylsulfonic acid.

10 Suitable odor controlling agents include copolymers of an acidic group containing monomer with one or more other monoethylenically unsaturated monomers. The above acidic monomers can be copolymerized, for example, with amides or nitriles of monoethylenically unsaturated carboxylic acids such as acrylamide, methacrylamide,

15 N-vinylformamide, N-vinylacetamide, N-methyl-vinylacetamide, acrylonitrile, methacrylonitrile. Other comonomers are vinylester of saturated C<sub>1</sub>- bis C<sub>4</sub>-carboxylic acids such as vinylformate, vinylacetate, vinylpropionate; alkyl vinylether compounds with at least 2 C-atoms within the alkyl-group such as ethyl vinylether,

20 butyl vinylether; esters of monoethylenically unsaturated C<sub>3</sub>- to C<sub>6</sub>-carboxylic acids such as esters from primary C<sub>1</sub>- to C<sub>18</sub>-alcohols and acrylic acid, methacrylic acid or maleic acid, halfesters of maleic acid such as maleic acid monomethylester, N-vinylactames such as N-vinylpyrrolidone or N-vinylcaprolactame,

25 acrylic acid- or methacrylic acid esters of alkoxylated primary saturated alcohols such as alcohols with 10 to 25 C-atoms which are reacted with 2 to 200 mol ethylenoxide and / or propylenoxide per mole of alcohol, monoacrylic acid esters and monomethacrylic acid esters of polyethylene glycol or polypropylene glycol up to

30 a molecular weight of 2000. Further monomers are styrene and alkylsubstituted styrene compounds such as ethyl styrene or tert.-butyl styrene.

The copolymers may contain the non-acidic monomers, for example, 35 in an amount of from 0 to 60 % by weight, preferably less than 20 % by weight.

Preferred polymers used as odor control means are homopolymers of acrylic acid, homopolymers of methacrylic acid, copolymers of 40 acrylic and methacrylic acid, copolymers of acrylic acid and maleic acid and copolymers of methacrylic acid and maleic acid. These polymers may have an average molecular weight Mw of from 1,000 to 5 million, preferably of from 1,500 to 150,000.

45 Examples of suitable odor controlling means are amphiphilic copolymers which contain units of

(a) hydrophobic monoethylenically unsaturated monomers and  
(b) monoethylenically unsaturated carboxylic acids, maleic anhydride, monoethylenically unsaturated sulfonic acids, monoethylenically unsaturated phosphonic acids or mixtures  
5 thereof.

These amphiphilic copolymers are prepared, for example, by polymerizing in an aqueous medium in the presence of at least one initiator at least one hydrophobic monoethylenically unsaturated monomer

(a) selected from the group consisting of styrene, methylstyrene, ethylstyrene, acrylonitrile, methacrylonitrile, C<sub>2</sub>- to C<sub>18</sub>-olefins, esters of monoethylenically unsaturated C<sub>3</sub>- to C<sub>5</sub>-carboxylic acids and monohydric alcohols, vinyl alkyl ethers, vinyl esters or mixtures thereof. From this group of monomers, isobutene, diisobutene, styrene and acrylic esters, such as ethyl acrylate, isopropyl acrylate, n-butyl acrylate, 20 and sec-butyl acrylate,

and at least one hydrophilic monomers

(b) selected from the group consisting of acrylic acid, methacrylic acid, maleic acid, maleic anhydride, itaconic acid, 25 vinylsulfonic acid, 2-acrylamidomethylpropanesulfonic acid, acrylamidopropane-3-sulfonic acid, 3-sulfopropyl acrylate, 3-sulfopropyl methacrylate, styrenesulfonic acid, vinylphosphonic acid or mixtures thereof in polymerized form.

30 If the amphiphilic copolymers are not sufficiently water-soluble in the form of the free acid, they are used in the form of water-soluble salts; for example, the corresponding alkali metal, alkaline earth metal and ammonium salts are used. These salts are 35 prepared, for example, by partial neutralization of the free acid groups of the amphiphilic copolymers with bases; for example, sodium hydroxide solution, potassium hydroxide solution, magnesium oxide, ammonia or amines, such as triethanolamine, ethanolamine, morpholine, triethylamine or butylamine, are used for the neutralization. Preferably, the acid groups of the amphiphilic copolymers are neutralized with sodium hydroxide or ammonia. The pH of the neutralized polymer solutions is for example less than 6.5, 40 preferably less than 5.0. The average molecular weight M<sub>w</sub> of the amphiphilic copolymers is, for example, from 1000 to 5,000,000, 45 preferably from 1500 to 150,000.

Particularly preferred amphiphilic copolymers contain

(a) from 95 to 45% by weight of isobutene, diisobutene, styrene or mixtures thereof and

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(b) from 5 to 55% by weight of acrylic acid, methacrylic acid, maleic acid, monoesters of maleic acid or mixtures thereof

as polymerized units. These copolymers may additionally contain

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(c) further monomers

as polymerized units. The copolymers can, if required, contain units of monoesters of maleic acid as polymerized further mono-

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mers (c). Such copolymers are obtainable, for example, by copolymerizing copolymers of styrene, diisobutene or isobutene or mixtures thereof with maleic anhydride in the absence of water and reacting the copolymers with alcohols after the polymerization, from 5 to 50 mol% of a monohydric alcohol being used per mole of anhydride groups in the copolymer. Suitable alcohols are, for example, methanol, ethanol, n-propanol, isopropanol, n-butanol, isobutanol and tert-butanol. However, it is also possible to react the anhydride groups of the copolymers with polyhydric alcohols, such as glycol or glycerol. Here, however, the reaction is continued only until only one OH group of the polyhydric alcohol reacts with the anhydride group. If the anhydride groups of the copolymers are not all reacted with alcohols, the anhydride groups not reacted with alcohols undergo ring opening as a result of the addition of water.

20

Other suitable odor controlling polymers may be obtained by grafting of synthetic or natural polymers with one or more of the above acidic monomers. Suitable backbones for grafting are, for example, starch, cellulose and derivatives thereof, polysaccharides and oligosaccharides, polyvinyl alcohol, polyalkylene oxides (preferably polyethylene oxide and /or polypropylene oxide), polyamines, polyamides, hydrophilic polyesters, galactomannans, guar derivatives and alginates. Grafted copolymers also include hydrolyzed starch-acrylonitrile graft copolymers, partially neutralized hydrolyzed starch-acrylonitrile graft copolymers, starch-acrylic acid graft copolymers, and partially neutralized starch-acrylic acid graft copolymers. Other suitable odor controlling agents are, for example, saponified vinyl acetate-acrylic ester copolymers, hydrolyzed acrylonitrile polymers and hydrolyzed acrylamide copolymers. These polymers may be used either

independently or in the form of a mixture with two or more other monomers.

The blend formed in step (a) contains at least one odor control means, for example, in an amount of from 0.01 to 30 % by weight. In the alternative it does not contain an odor control means which is then added to the coated fabric after polymerization. The effective concentration range varies widely depending on the mechanism of the odor control means, for example, antimicrobials are used in amounts of from 0.01 to 1 % by weight, whereas the amounts of inorganic materials, pH-controlling materials and chelants are usually in the range of from 1 to 30 % by weight.

#### Skin care agents

15 Skin care agents are known in the art. Examples are camomile extract, aloe vera, panthotenate and esters.

#### Initiators

20 The blend formed in step (a) contains one or more initiators. Suitable initiators include the initiators and initiator combinations described above as being useful in the production of superabsorbent polymer particles. In addition, it may be desirable to 25 use initiators designed to decompose when subjected to ultraviolet light and/or electron-beam ("e-beam") irradiation. Preferred initiators include water-soluble azo compounds such as 2,2'-azo-bis(2-(2-imidazole-2-yl))propane dihydrochloride and 2,2'-azo-bis(amidino)propane dihydrochloride, water soluble benzophenones 30 such as 4-benzoyl-N,N,N-trimethylbenzenemethanaminium chloride, 2-hydroxy-3-(4-benzoylphenoxy)-3-N,N,N-trimethyl-1-propanaminium chloride monohydrate, 2-hydroxy-3-(3,4-dimethyl-9-oxo-9H-thioxanthone-2-yloxy-N,N,N-trimethyl-1-propanaminium chloride, 2-hydroxy-1-[4-(hydroxyethoxy)phenyl]-2-methyl-1-propanone, 35 2-hydroxy-2-methyl-1-phenyl-propan-1-one, and 4-benzoyl-N,N-dimethyl-N-[2-(1-oxo-2-propenyl)oxy]ethylbenzenemethanaminium chloride. In general, the sprayable blend contains one or more initiators at a level sufficient to initiate polymerization of the superabsorbent-forming monomer in the blend. The blend contains one or 40 more initiators at a level sufficient to result in the complete polymerization of the superabsorbent polymer forming monomer in the said blend, generally at a level within the range of from 0.01 to 5.0, most preferably at a level from 0.2 to 2.0 percent by weight of superabsorbent-forming monomer in the blend. When 45 using a combination of initiators in the blend, such as redox package, it is possible to incorporate one of the initiators, such as the reducing agent, into the blend along with the other

components of the blend and incorporate one or more additional initiators, such as an oxidizing agent, into the blend just before the sprayable blend exits, for example, the nozzle of a spray equipment which is used to spray the sprayable blend onto the fibrous web. A particularly preferred combination of initiators includes both an azo initiator and 2-hydroxy-1-[4,(hydroxyethoxy)phenyl]-2-methyl-1-propanone.

#### Water

In addition to the superabsorbent polymer particles, monomers forming superabsorbent polymers and one or more initiators, the blend contains water. Generally, the blend contains sufficient water to render the Brookfield viscosity (measured at 20°C, 20 rpm, spindle 02) of the blend of at least 20 mPas. For example, the viscosity is in the range of from 20 to 1,000 mPas, preferably 20 to 400 mPas. The level of water in the blend is generally in the range from 40 to 80 percent by weight, more preferably from 50 to 60 percent by weight of the blend. Blends having a viscosity of up to 400 mPas are preferred, because they can be easily sprayed onto both sides of the web by spraying through a nozzle.

#### Softening agent

If desired, a softening agent may be added to the foregoing components of the blend formed in step (a). The amount of softening agent used in the blend may be of from 0.2 to 20 % by weight, preferably 5 to 15 % by weight. Softening agents include such compounds as quaternary ammonium compounds (e.g. di(hydrogenated) tallow dimethylammonium chloride, di(hydrogenated) tallow dimethylammonium methyl sulphate) and / or polyhydroxy compounds selected from glycerol, polyglycerols, polyoxyethylene glycols, polyoxypropylene glycols, or mixtures thereof. If a softening agent is added to the blend, glycerol is preferably used as softening agent.

#### Crosslinker

It is generally desirable to include in the blend formed in step (a) one or more crosslinkers. It is particularly preferred to use one or more crosslinkers in this blend, for example thermal, redox and / or UV initiators. Suitable crosslinkers include those described above for the preparation of the superabsorbent polymer particles. Preferred crosslinkers include ethoxylated and propoxylated trimethylolpropane triacrylate derivatives such as SR-9035 and SR-492 available from Sartomer Co., Inc. of Exton, PA. When

used, the crosslinkers are present in the blend formed in step (a), for example, in an amount of from 0.05 to 5.0, preferably 0.1 to 1 % by weight, based on the weight of the monomers forming the superabsorbent polymer.

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#### The blend

In the process a blend is formed in step (a) by combining monomer forming superabsorbent polymers, superabsorbent polymer particles, optionally odor control means or mixtures thereof, water, optionally softening agents or mixtures thereof and initiator. The blends preferably contain a softening agent. The most preferred softening agent is glycerol. Although the order of combining these materials is not particularly important, for safety 15 reasons it is preferred to add the initiator last. The amounts of the individual components of the blend are set forth above and are generally selected so that the Brookfield viscosity (measured at 20°C, 20 rpm, spindle 02) is at least 20 mPas. The blend formed in step (a) may have, for example, a viscosity of from 20 to 20,000 mPas, preferably of from 20 to 400 mPas, more preferably from 30 to 150 mPas, most preferably from 40 to 100 mPas. Many factors will influence the viscosity of the blend, including the chemical nature and size of the superabsorbent polymer particles and odor control means, the extent of neutralization of the superabsorbent polymer particles, the extent of neutralization of the one or more superabsorbent-polymer-forming monomers, and the concentration of the superabsorbent polymer particles.

After the blend has been prepared, it is applied onto both sides 30 of a fibrous web in step (b1). The blend can be printed onto a web and is preferably sprayed in step (b1) onto a fibrous web. As used herein, spraying is intended to include any suitable means for generating and delivering droplets of liquid. The spraying can be achieved by any conventional spray equipment. The equipment can be airless, air-assisted airless or can utilize pressurized air. One or more inert gases, such as nitrogen, argon or helium, may be substituted for some or all of the air to assist in the removal of oxygen from the sprayable blend during the spraying process.

40

The loading level of the blend applied to the fibrous web is highly dependent on the application of the absorbent article. The blend is applied to the fibrous web, for example, in amounts of from 20 to 800 g/m<sup>2</sup>, with reference to the solids including the monomers. Preferably of from 40 to 300 g/m<sup>2</sup> and most preferably of from 60 to 150 g/m<sup>2</sup> of the solids including the monomers of the blend are applied onto the web. For example, a blend having a

solids content of 50 % by weight is applied to a fibrous web in an amount of from 40 to 1,600 g/m<sup>2</sup>. If a spray equipment is used for the application of the blend, it should be adjusted to vary the droplet size of the spray taking into account such factors as 5 the size of the superabsorbent polymer particles in the sprayable blend and the desired particle size of superabsorbent polymer particles on the final product.

Alternatively, after the blend has been prepared, it is applied 10 onto a fibrous web in step (b2). The blend can be printed onto a web and is preferably sprayed in step (b2) onto a fibrous web. As used herein, spraying is intended to include any suitable means for generating and delivering droplets of liquid. The spraying can be achieved by any conventional spray equipment. The equipment 15 can be airless, air-assisted airless or can utilize pressurized air. One or more inert gases, such as nitrogen, argon or helium, may be substituted for some or all of the air to assist in the removal of oxygen from the sprayable blend during the spraying process.

20 The loading level of the blend applied to the fibrous web is highly dependent on the application of the absorbent article. For example, a blend having a solids content of 46 % by weight is applied to a fibrous web in an amount of from 20 to 1,500 g/m<sup>2</sup>. If 25 the absorbent article is used as an absorbent core of a hygiene article the amount of the blend having a concentration of 46 % by weight is, for example, of from 1,000 to 1,500 g/m<sup>2</sup>. Blends having another solids content are used in such amounts that a corresponding solids content is achieved on the coated web. The blend is 30 preferably applied by spraying to the fibrous web. For example, a sprayable blend having a viscosity of from 20 to 400 mPas (measured at 20°C, 20 rpm, spindle 02) and having a solids content of 46 % by weight is sprayed onto the web in amounts of up to 1,500, preferably of from 20 to 800 g/m<sup>2</sup> and most preferably of from 60 35 to 300 g/m<sup>2</sup>. The spray equipment should be adjusted to vary the droplet size of the spray taking into account such factors as the size of the superabsorbent polymer particles in the sprayable blend and the desired particle size of superabsorbent polymer particles on the final product.

40 The blend can be applied homogeneously to a web on one or both sides or can also be applied to it in a pattern which corresponds to a desired distribution. For example, the edges of the web may have a higher loading than the other parts of the web or it may 45 contain a homogeneous coating having several strips of higher loadings.

In order to apply different loadings of a sprayable blend on a fibrous web, there are, for example, several possibilities:

- a) changing the spraying angle by electronical variation of the opening of the spray nozzle for different loadings in x/y direction
- 5 b) changing the spraying angle by variation of the distance between spray nozzle and web for different loadings in x/y direction
- 10 c) raising the amount of sprayable blend at certain regions of the web for different loadings in z direction. The continuous feed to the inlet tubes (spray nozzles) is changed at certain intervals by raising the flow velocity of the sprayable blend in such a way that the resulting loading of the web is increased.
- 15 d) stopping the web at certain intervals during the continuous process for different loadings in z direction. The constant web transport velocity is interrupted or slowed down at certain intervals.

20 The spray equipment should be adjusted to vary the droplet size of the spray taking into account such factors as the size of the superabsorbent polymer particles in the sprayable blend and the desired particle size of superabsorbent polymer particles in the 25 final product.

For example, in regions of higher loading, a sprayable blend having a solids content of 46 % by weight is sprayed onto a fibrous web in an amount of from 200 to 1,200 g/m<sup>2</sup>, preferably of from 250 30 to 800 g/m<sup>2</sup>, most preferably from 300 to 500 g/m<sup>2</sup>. These loading are representative for the preparation of absorptive cores used in hygiene articles.

In a preferred embodiment of the invention the blend is applied 35 to the fibrous web in such a way that the superabsorbent particles are uniformly distributed on the fibrous web.

#### The Web

40 Suitable fibrous webs for the present invention include those made using synthetic polymeric fibers. The synthetic polymeric fibers may be formed from any polymeric material capable of forming fibers which fibers can be formed into a fibrous web. Suitable polymeric material from which the synthetic polymeric fibers 45 may be formed include polyolefins, such as polyethylene, polypropylene, and the like; polyesters such as polyethylene terephthalate and the like; polyamides such as nylon 6, nylon 6,6,

poly(iminocarboxylpentamethylene) and the like; acrylics, and modified cellulosic material, such as cellulose acetate and rayon; as well as mixtures and copolymers thereof.

5 The synthetic polymeric fibers may be formed by meltblowing, through a spunbond process, by extrusion and drawing, or other wet, dry and melt spinning methods known to those skilled in the art. The synthetic polymeric fibers from which the web is formed may have a discrete length or may be substantially continuous.

10 For example, if the synthetic polymeric fibers are formed by meltblowing, the fibers may be substantially continuous (few visible ends). If the fibers are formed by extrusion and drawing to produce a tow, the tow may be used as produced or cut into staple fibers having a length, for example of from about 25 millimeters

15 to about 75 millimeters or short cut into length of from about 1 millimeter to about 25 millimeters. The synthetic polymeric fibers may suitably have a maximum cross-sectional dimension of from about 0.5 micrometer to about 50 micrometers as determined by microscopic measurement using an optical microscope and a calibrated stage micrometer or by measurement from Scanning Electron photomicrographs.

The fibrous web may be formed directly through a spunbond or meltblown process, or by carding or air-laying staple or short cut fibers. Other methods of forming fibrous webs known to those skilled in the art may be suited for use in the present invention. The web may subsequently be bonded to enhance structural integrity. Methods of bonding fibrous webs are known to those skilled in the art and include thermal bonding, point bonding, powder bonding, ultrasonic bonding, chemical bonding, mechanical entanglement, and the like. The fibers may be homogenous fibers or may be a core/sheath or side-by-side fibers known in the art as bicomponent fibers.

35 The fibrous web may be formed from a single type of synthetic polymeric fiber or may contain synthetic polymeric fibers formed from different polymeric materials, having different fiber lengths or maximum cross-sectional dimensions. For example, the web may comprise a mixture of (1) bicomponent fibers having a

40 polyethylene sheath and a polypropylene core which bicomponent fibers have a maximum cross-sectional dimension of about 20 micrometers and a length of about 38 millimeters and (2) polyester fibers (polyethylene terephthalate) having a maximum cross-sectional dimension of about 25 micrometers and a length of about 38

45 millimeters. Fibers 1 and 2 may be combined in a weight ratio of

from 1:99 to 99:1. The fibers may be uniformly mixed or may be concentrated at opposite planar surfaces of the fibrous web.

The web suitably comprises from about 10 to 100 weight percent, 5 beneficially of from about 20 to 100 weight percent, preferably of from about 25 to 100 weight percent, and most preferably of from about 50 to 100 weight percent synthetic polymeric fibers. In addition to the synthetic polymeric fibers, the web may contain from about 90 to 0 weight percent of a nonsynthetic poly- 10 meric fiber such as wood pulp fluff, cotton liners, cotton, and the like.

In one preferred embodiment, the web contains synthetic polymeric fibers which are formed from a polymeric having a high wet modulus. The importance of the modulus of a material is discussed in 15 the book "Absorbency" edited by P.K. Chatterjee (Elsevier, N.Y., 1985). A polymeric material will be considered to have a high wet modulus when it has a wet modulus greater than about 80 percent of its dry modulus as determined by ASTM (American Society for 20 Testing and Materials) test method D 2101-91 using modified gauge lengths. It is often desired to form the synthetic polymeric fibers of the web from a polymeric material having a high wet modulus because such material generally form fibrous webs which possess a relatively high degree of wet resiliency. The wet resilience 25 of a fibrous web is related to the pore structure (while under load) of the fibrous web. As will be discussed in greater detail below, it is often desired that the web have a relatively high degree of wet resilience.

30 The pore structure (while under load) of a fibrous structure formed from fibers of a polymeric material will, as discussed above, relate to the wet and / or dry modulus of the constituent fibers. Wet modulus of the constituent fibers should be considered for fibers that may likely be wetted during use. For the 35 purposes of estimating the effect of load on the pore structure of a fibrous structure formed from fibers of a polymeric material the tensile modulus of the fiber which can be related to the flexural rigidity of the fiber as shown in the book "Physical Properties of Textile Fibers" by W. E. Morton and J. W. S. Hearl 40 (The Textile Institute, London, 1975) can be used.

As a general rule, the polymeric materials from which the synthetic polymeric fibers of the web are formed will be inherently hydrophobic. As used herein, the term "hydrophobic" describes a 45 material which has a contact angle of water-in-air of greater than 90 degrees. The term "hydrophilic" refers to a material which has a water-in-air contact angle of less than 90 degrees.

The water-in-air contact angle is suitably determined as set forth in the book "Absorbency" edited by P.K-Chatterjee (Elsevier, N.Y., 1985). As used herein, a polymeric material will be considered as "inherently" hydrophobic or hydrophilic when the 5 polymeric material, free from any surface modifications or treatments, e.g., surface active agent, spin fishes, blooming agents, etc., is hydrophobic or hydrophilic, respectively.

When the synthetic polymer fibers of the web are formed from a 10 polymeric material which is inherently hydrophobic, it is often desirable to treat the fibers with a surface modifying material to render the surface of the fiber hydrophilic. For example, a surfactant may be applied to the fibers.

15 The web suitably has a basis weight of from about 20 to about 200 g/m<sup>2</sup>, beneficially of from about 50 to about 150 g/m<sup>2</sup>, and preferably of from about 25 to about 125 g/m<sup>2</sup>.

20 The web suitably has a density of from about 0.005 to about 0.20 g/cm<sup>3</sup>, beneficially of from about 0.01 to about 0.16 g/cm<sup>3</sup>, and preferably of from about 0.08 to about 0.14 g/cm<sup>3</sup>.

25 The fibrous web may also comprise hydrophilic fibers. The hydrophilic materials may be inherently hydrophilic such as cellulosic fibers such as wood pulp fluff, cotton linters, and the like; regenerated cellulose fibers such as rayon; or certain nylon copolymers such as poly(pentamethylenecarbonamide) (nylon-6)/polyethylene oxide. Alternatively, the hydrophilic fibers may be obtained from hydrophobic fibers by treatment with a hydrophilizing 30 agent. For example, the fibers may be formed from a polyolefin material which is subsequently coated with a surface active agent such that the fiber itself is hydrophilic as described herein. Other methods of hydrophilizing fibers formed from hydrophobic materials are known and suited for use in the present invention.

35 Methods of providing inherently hydrophilic fibers such as wood pulp fluff are known. If the hydrophilic fibers are obtained from hydrophobic fibers which have been treated to possess a hydrophilic surface, the fibers will suitably have a fiber length and 40 a maximum cross-sectional dimension as set forth above. If the hydrophilic fibers are inherently hydrophilic such as wood pulp fluff, rayon, cotton, cotton linters and the like, the fibers will generally have a length of from about 1.0 millimeters to about 50 millimeters and a maximum cross-sectional dimension of 45 from about 0.5 micrometers to about 100 micrometers.

The fibrous web suitably comprises from about 10 to 100 weight percent, beneficially from about 30 to 100 weight percent, and preferably from about 55 to 100 weight percent of hydrophilic fibers, preferably inherently hydrophilic fibers. In addition to 5 the hydrophilic fibers, the web may contain from about 90 to 0 weight percent of high wet modulus, preferably inherently hydrophobic fibers. The web may be formed from a single type of hydrophilic fiber or may contain hydrophilic fibers having different compositions, lengths and maximum cross-sectional dimensions.

10 In one preferred embodiment, the web is formed from air-laid cellulosic fibers such as wood pulp fluff. Wood pulp fluff fibers are preferred for use due to their ready availability and due to the fact that the fibers are relatively inexpensive compared to 15 synthetic fibers.

In one especially preferred embodiment, the web is compressed under reduced pressure.

20 The compressed web suitably has a basis weight of from about 40 to about 400 g/m<sup>2</sup>, beneficially of from about 60 to about 300 g/m<sup>2</sup>, and preferably of from about 50 to about 200 g/m<sup>2</sup>.

25 The compressed web suitably has a density of from about 0.15 to about 0.40 g/cm<sup>3</sup>, beneficially of from about 0.12 to about 0.30 g/cm<sup>3</sup>, and preferably of from about 0.10 to about 0.20 g/cm<sup>3</sup>.

It is a preferred embodiment of the present invention to provide 30 profiled absorbent structures of at least two uniformly coated webs in which the characteristics of each individual web have been optimized to perform a specific function such as acquisition, fluid distribution or fluid storage. An example of such an optimized structure would be a two layered structure in which one coated web has been optimized for fluid acquisition and the other 35 for fluid storage. The performance characteristics of spray coated webs are dependent on a number of different factors such as the x-link density and degree of loading of the formed superabsorbent polymer as well as the density and basis weight of the substrate itself.

40 Spray coated webs optimized for fluid acquisition are preferably prepared from webs with a basis weight of 10 to 80 g/m<sup>2</sup>, more preferably 20 to 60 g/m<sup>2</sup> with a density in the range of 0.005 to 0.08 g/cm<sup>3</sup>. The amount of applied superabsorbent polymer loading is 45 preferably in the range of 10 to 80 percent by weight, more pre-

ferably 20 to 70 % by weight and most preferably 30 to less than 65 % by weight.

It is further desired to use a highly crosslinked superabsorbent 5 coating for acquisition layers to both minimize fluid retention and to increase swelling speed. Crosslinker amounts are preferably between 0.1 to 5.0%, more preferably between 0.3 to 4.0 % and most preferably between 0.5 to 3.0 % based on the amount of reactive monomer in the sprayable blend.

10

Spray coated webs optimized for fluid storage are preferably prepared from webs with a basis weight of 10 to 200 g/m<sup>2</sup>, more preferably 40 to 150 g/m<sup>2</sup> with a density in the range of 0.005 to 0.08 g/cm<sup>3</sup>. The amount of applied superabsorbent polymer loading 15 is preferably in the range of 10 to 99 percent by weight, more preferably 20 to 90 % by weight and most preferably 30 to 85 % by weight.

It is further desired to use a lightly crosslinked superabsorbent 20 coating for storage layers to maximize absorptive capacity.

Crosslinker amounts are preferably between 0.1 to 1.0%, based on the amount of reactive monomer in the sprayable blend.

Another desirable feature of the present invention is the incorporation of wicking layers in combination with webs coated with a sprayable blend to enhance fluid distribution within the described profiled absorbent structures. These distribution layers not only provide fluid transport in the x-y plane of the absorbent article, which promotes better utilization of the total available 30 absorptive capacity of the absorbent article, but also provides drainage of the pore volume of the hydrated spray coated webs, which yields improved surface dryness as measured by rewet tests.

#### Curing/Polymerizing

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After the blend has been applied onto the fibrous web, this composite is subjected to conditions under which the monomer forming the superabsorbent polymer will polymerize. Depending upon the type of initiator used in the blend, these conditions may include, for example, subjecting the fibrous web to which the blend 40 was applied to heat, ultraviolet radiation, e-beam radiation, or a combination thereof. Furthermore, the composite can be subjected to static or continuous conditions such as by moving the composite along a conveyor through regions of radiation or heat.

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For thermal curing there are no particular limitations on the type of reaction vessel used. For batch polymerizations, sprayed webs may be cured in an oven in an air or inert atmosphere, and optionally under vacuum. In the case of a continuous process, the 5 web may be passed through a dryer, such as an infrared ("IR"), through air or the like. The polymerization temperature can vary depending on the thickness of the substrate, the concentration of monomer and the type and amount of thermal initiator used in the blend. The polymerization is typically carried out in the temper- 10 ature range, for example, of from 20°C to 150°C and preferably of from 40°C to 100°C. The polymerization time depends on the polymerization temperature, but is typically several seconds to 2 hours and preferably several seconds to 10 minutes. After polymerization is completed, the web can then be dried to the desired 15 moisture content.

UV curing of webs coated with the sprayable blends may be conducted by the use of a conventional UV lamp. The conditions under which the irradiation is conducted, such as radiation intensity 20 and time may differ depending on the type of fibrous substrate used, the amount of monomer applied to the substrate and the like. However, irradiation is generally conducted using a UV lamp with an intensity in the range of from 100 to 700 watts per inch ("W/in"), preferably in the range of from 400 to 600 W/in for 0.1 25 seconds to 10 minutes, with the distance between the UV lamp and the substrate being 2 to 30 centimeters. The irradiation of the composite with ultraviolet rays may be conducted under vacuum, in the presence of an inorganic gas, such as nitrogen, argon, helium and the like, or in air.

30 The temperature during irradiation is not critical, and the irradiation of the sprayed web can be satisfactorily conducted at room temperature.

35 Electron beam curing can be accomplished using a commercially available electron beam accelerator, such as the Electrocurtain® CB 175 (Energy Sciences, Inc., Wilmington, MA). Accelerators operating in the 150 to 300 kilovolt range are acceptable. The beam current on such systems, typically 1 to 10 milliamperes, can be 40 adjusted to obtain the desired dose of ionizing radiation. The ionizing radiation dose employed will vary somewhat, depending on factors such as the presence or absence of crosslinking monomers, desired degree of polymerization of the polymer, degree of crosslinking desired, and the like. In general, it is desirable to ir- 45 radiate the coated web with doses from about 1 to 16 megarads, more preferably 2 to 8 megarads. Particularly when using lower doses, it is desirable to purge oxygen from the sprayable blend

(as by bubbling nitrogen through the solution). The maximum dose would be that dose at which degradation of the fibers begins.

After irradiation, the coated web may be dried to remove water by 5 such means as forced air ovens, infrared lamps and the like.

#### Storage Layer

The absorbent structures according to the present invention are 10 suitable for use in disposable absorbent products such as diapers, training pants, adult incontinence products, feminine care products, wound dressings and the like. Methods of forming such absorbent products and the absorbent products formed thereby are known to those skilled in the art and are described, for example, 15 in the following U.S. Pat. Nos.: 4,944,735 issued Jul. 31, 1990 to Mokry; 4,798,603 issued Jan. 17, 1989, to Meyer et al.; 4,710,187 issued Dec. 1, 1987, to Boland et al.; 4,770,656 issued Sep. 13, 1988, to Proxmire et al.; and 4,762,521 issued Aug. 9, 1988, to Roessler et al.; the disclosures of which are incorporated 20 herein to the extent they are consistent herewith. The absorbent articles according to the invention can be used in disposable hygiene products as storage layer, wicking layer and acquisition layer. They can be designed in such a way that they can substitute the absorbent core the storage layer and storage and 25 acquisition layer in a conventional disposable hygiene article.

The absorbent structures of the present invention suitably form a storage layer or combined storage acquisition layer (dual layer) of a disposable absorbent product. Such an storage or dual layer 30 core is suitably sandwiched between, and in liquid communication with, a bodyside liner (also known as a top sheet), and a liquid impervious or relatively liquid impervious outer cover. In order to function well as a storage layer or dual layer, an absorbent structure should exhibit rapid uptake of fluid, good transfer 35 properties, good uptake upon repeated insults with fluid, good skin compatibility, low rewet and high capacity on storage.

The absorbent structures of the present invention exhibit these properties and also have further advantages. Because the absorbent structures of the present invention can be prepared with superabsorbent polymer particles that are at least partially neutralized, and because the superabsorbent forming monomer used in the sprayable blend can be at least partially neutralized, it is possible to control the pH of the resulting absorbent structure 45 formed by polymerizing the sprayable blend sprayed onto the fibrous web. By controlling the pH of the absorbent structure, particularly within the range of from about 4.3 to about 5.5,

several advantages can result. For example, an absorbent structure having a pH within that range should be compatible with skin, should exhibit reduced bacterial growth, should reduce fecal proteolytic activity and lipolytic enzymatic activity, should 5 control odor and contain ammonia. The benefits associated with the control of pH in a top sheet of a disposable hygienic article are discussed, for example, in U.S. Patent No. 4,657,537, the disclosure of which is hereby incorporated by reference. Similar benefits are expected to result from an acquisition layer having 10 a pH in the range of from about 4.3 to about 5.5. Although the partial neutralization of the superabsorbent polymer particles and superabsorbent forming monomer will tend to reduce the overall absorptive capacity of the absorbent structure of the present invention, total absorptive capacity is not the only critical 15 feature when using the structures as a dual layer in a disposable hygiene product.

It is further believed that the absorbent structures of the present invention perform well as storage and dual layers in a disposable diaper because the swelling of the superabsorbent polymer 20 particles is capable of expanding the fibrous web, particularly if the web were compressed, so that the interstitial pore volume of the web increases after an insult of liquid. This increase in the interstitial pore volume of the web contributes to the rapid 25 uptake of fluids in the web. Accordingly, enhancing the speed at which the interstitial pore volume is generated, such as by the rapid swelling of the superabsorbent polymer particles, further contributes to the rapid uptake of fluids in the web. The free-swell expansion volume ("FSEV") test and the expansion volume under load ("EVUL") test described below are an indirect measure of 30 the rate of formation of the interstitial pore volume in the web. In general, it is desirable to maximize the values of the FSEV and EVUL in designing a high performing superabsorbent article. At a minimum the absorbent structures of the present invention 35 have free-swell expansion volumes and/or expansion volumes under load of at least about 0.1 milliliters within about 5 minutes, preferably within about 30 seconds, and most preferably within about 5 seconds. Preferably, the absorbent structures of the present invention have free-swell expansion volumes of at least about 0.5 milliliters within about 5 minutes, preferably within 40 about 30 seconds, and most preferably within about 5 seconds. Most preferably, the absorbent structures of the present invention generally have free-swell expansion volumes of at least about 1.0 milliliters within about 5 minutes, preferably within 45 about 30 seconds, and most preferably within about 5 seconds. Preferably, the absorbent structures of the present invention generally have expansion volumes under load of at least about 0.2

milliliters within about 5 minutes, preferably within about 30 seconds, and most preferably within about 5 seconds. Most preferably, the absorbent structures of the present invention generally have expansion volumes under load of at least about 0.3 milliliters within about 5 minutes, preferably within about 30 seconds, and most preferably within about 5 seconds.

The dual layer may have an FSEV of at least about 0.1 ml within 5 minutes to about 1.0 ml within about 5 seconds and an EVUL of at least about 0.1 ml within about 5 minutes to about 0.3 ml within about 5 seconds.

The absorbent core comprises a storage layer or a dual layer comprising a separate storage layer or a storage layer function. The storage layer or storage layer function is present in the absorbent structure in an amount of from about 50 to about 100, preferably of from about 60 to about 95, and most preferably of from about 70 to about 90 weight percent based on total weight of the absorbent core. If an additional acquisition layer is used, it is present in an amount of about 50 or less, preferably from 5 to 40, most preferably from 10 to 30 percent based on the total weight of the absorbent core.

Means of containing acquisition layer are known to those skilled in the art. Any means of acquisition layer is suited for use in the present invention. The storage layer or dual layer may be combined with other high absorbency material.

As used herein, the term "high-absorbency material" refers to a water-swellable, generally water-insoluble material capable of absorbing at least about 10, desirably about 20, and preferably about 50 times or more its weight in water. The high-absorbency material may be formed from organic material, which may include natural materials such as agar, pectin, and guar gum, as well as synthetic materials such as synthetic hydrogel polymers. Synthetic hydrogel polymers include, for example, carboxymethyl cellulose, alkali metal salts of polyacrylic acid, polyacrylamides, polyvinyl alcohol, ethylene maleic anhydride copolymers, polyvinyl ethers, hydroxypropyl cellulose, polyvinyl morpholinone, polymers and copolymers of vinyl sulfonic acid, polyacrylates, polyacrylamides, polyvinyl pyridine, polyvinylamines, and the like. Other suitable polymers include hydrolyzed acrylonitrile grafted starch, acrylic acid grafted starch, and isobutylene maleic anhydride copolymers and mixtures thereof. The hydrogel polymers are preferably lightly crosslinked to render the materials substantially water insoluble. Crosslinking may, for example, be by irradiation or covalent, ionic, van der Waals, or hydrogen

bonding. Suitable materials are available from various commercial vendors such as the Dow Chemical Company, BASF Corporation, Nippon Shokubai., and Stockhausen Inc. The high-absorbency material may be in the form of particles, spheres, flakes, fibers, rods, 5 films or any of a number of geometric forms. When in the form of particles or spheres, it may be desired that the particles or spheres have a maximum cross-sectional dimension of from about 10 micrometers to about 2000 micrometers, preferably from about 60 micrometers to about 1000 micrometers.

10 In one embodiment it is desired that the high-absorbency material have the ability to absorb a liquid while under a load. The ability of a high-absorbency material to absorb a liquid while under a load is quantified as the Absorbency Against Pressure (AAP) 15 value. Specifically, the AAP value is the amount (in grams) of an aqueous solution containing 0.9 weight percent sodium chloride, a gram of the high-absorbency material can absorb in 60 minutes under a load of 0.3 pound per square inch. As a general rule, it is desired that the high-absorbency material have an AAP value of at 20 least about 10, desirably at least about 15, and preferably at least about 25.

The absorbent core suitably has a basis weight of from about 200 to about 1000 g/m<sup>2</sup>, beneficially of from about 250 to about 750 25 g/m<sup>2</sup>, and preferably of from about 300 to about 500 g/m<sup>2</sup>. The absorbent layer suitably has a density of from about 0.06 to about 0.40 g/cm<sup>3</sup>, beneficially of from about 0.12 to about 0.35 g/cm<sup>3</sup>, and most preferably of from about 0.15 to about 0.30 g/cm<sup>3</sup>.

30 A further advantage of locating the dual layer between the body of a wearer and the absorbent layer is that the acquisition layer, may have a relatively dry feel even after it has been wetted. This is because the acquisition layer includes synthetic polymeric fibers, is resilient and may be more easily desorbed by 35 the absorbent layer. Thus, a relatively dry surface may be presented for contacting a wearer's skin. In contrast, the absorbent layer including cellulosic or other inherently hydrophilic fibers can have a relatively wet feel. This wet surface is located remote from the body of a wearer and is spaced therefrom by the acquisition layer.

It is well known to those in the art of disposable hygienic products that the insertion of thick, lofty fabric structures between the topsheet and the absorbent core aids in the rate of uptake of 45 fluid insults from the surface of the article. U.S. Patent No. 5,364,382 discloses a number of key properties of these acquisition layers, such as the wet and/or dry modulus of the constitu-

ent fibers, the hydrophilicity of the fibers and resiliency of the fabric structure, that contribute to the ability of these materials to rapidly uptake fluids in an absorbent article. Such properties contribute to the acquisition layer's ability to: stay 5 open under load, maintain void volume, resist collapse when wetted, enhance the desorption properties of the fabric, and preserve void volume capacity after successive insults of fluid.

One advantage of the absorbent articles of the present invention 10 is the rate of formation of void volume following compression of the absorbent article. In many current, commercially manufactured absorbent products a considerable amount of pressure is applied during manufacture to produce an "ultrathin" product. Acquisition and storage materials used in these current products have no real 15 mechanism to reopen after being compressed other than the memory effects preserved within the fibers themselves. The incorporation of discrete superabsorbent particles into a fibrous web, as provided in the present invention, provides such a mechanism through the swelling of the superabsorbent particles following an insult 20 of fluid. Two important performance parameters of an acquisition and storage layer in a disposable hygienic article are (1) the degree of expansion and (2) the rate of expansion. Both of these parameters are indirectly measured by FSEV and EVUL. The degree of expansion is an indication of the pore volume available for 25 fluid uptake (i.e. larger volume correlates with better performance) and by increasing the speed at which this pore volume is generated the likelihood of leakage upon insult is diminished. Both of these properties maybe be influenced by adjusting such 30 parameters as: the degree loading of superabsorbent polymer particles on the web, the particle size of the superabsorbent polymer particles, the degree of swelling of the superabsorbent polymer particles, the degree of neutralization of the superabsorbent forming monomer in the sprayable blend, the degree of crosslinking and the like.

35 It has also been observed that webs, after being sprayed with a sprayable blend and subjected polymerization conditions, have other beneficial properties. As mentioned above, certain web materials are subjected to compression at one or more times during 40 the construction of a disposable hygienic article such as a diaper. After a web material has been compressed, there is a tendency for the fibers to relax, and expand somewhat thereby increasing the thickness of the web. However, this relaxation phenomenon is much less pronounced in articles prepared in accordance with the present invention which tend to remain stably in a 45 compact state until subjected to an insult of fluid.

Rewet and strikethrough testing of absorbent articles are common quality assurance tests used in the hygiene industry to measure surface dryness and the rate of fluid uptake, respectively, following successive fluid insults; therefore, these tests are useful for the evaluation of the performance of acquisition materials in absorbent products. A common undesirable trend seen among most commercially available diapers on the market today is the fact that strikethrough times tend to increase with successive doses of fluid during rewet testing. With conventional fluff-based absorbent structures the cellulosic fibers can lose resiliency and collapse when wetted. As a result, the liquid uptake rate of the wetted structures may become too low to adequately accommodate subsequent, successive fluid insults. Where absorbent gelling particles are incorporated between the fibers to hold them apart, the gelling particles swell and do not release the fluid. Swelling of the particles can then diminish the void volume of the absorbent structure and reduce the ability of the structure to rapidly uptake fluid. The degree to which the swelling of the absorbent gelling particle negatively impacts the rate of fluid uptake is dependent upon a number of factors such as the concentration of superabsorbent used in the absorbent core, the degree of cross-linking, the uniformity of the distribution of SAP within the structure, the particle size distribution, the hydrophobicity of the particle and the like. Each of these factors are easily controlled with the present invention and may be optimized to achieve the desired performance properties for a given absorbent article, particularly when used as an acquisition layer in an absorbent article.

In general, it can be seen that these acquisition and storage materials minimize or eliminate the trend of increasing strikethrough times with successive insults of fluid. This desired beneficial effect may be controlled and optimized with the present invention through the control of such parameters as: the concentration of superabsorbent polymer particles applied to the web structure, the particle size distribution of the resulting superabsorbent polymer particles, the rate of swelling of the particles, the degree of swelling of the particles and the like. A further observed beneficial effect is the reduction in rewet values. This effect may also be controlled through the above described parameters. In addition, it is further believed that lowering the degree of neutralization of the superabsorbent particles formed on the web, thereby increasing the hydrophobicity of the particles, further enhances this effect by increasing it's tendency to be drained by the underlying wood fluff/pulp/superabsorbent polymer absorbent core.

It is well known in the art that a hygienic absorbent article capable of lowering skin pH within the range of 3.0 to 5.5 is beneficial in preventing or at least reducing the incidence of diaper rash. Articles, compositions and procedures which inherently tend 5 to lower the pH of diaper skin are also known in the art. U.S. Patents 4,657,537, 4,382,919, 3,964,486, 3,707,148 and 3,794,034 teach the addition of various acidic pH control agents to absorbent articles or to the diaper skin environment. In those instances wherein acidic pH control agents have been incorporated 10 into the cores of the absorbent articles, significant amounts of acids are needed to bring about the desired absorption of ammonia or lowering of skin pH. Such approaches suffer from a number of drawbacks including: decreasing the absorptive capacity of the absorbent core, safety and comfort factors associated with leaching 15 of the materials from the article and processing problems associated with the placement and distribution of the acidic material within the absorbent core. U.S. Patent 4,657,537 discloses the preparation and use of topsheet materials containing ion exchange functionalities capable of controlling skin pH in urine 20 soiled baby diapers, however, the ion-exchange capacities of these materials are limited to the range of 0.25 to 1.0 meq./gram. Acquisition materials produced by the present invention may be prepared to contain from 1.0 to 10 meq./gram of ion-exchanging functionalities.

25

#### Test methods

The centrifuge retention capacity ("CRC") is a measure of the amount of fluid retained after being centrifuged. The CRC of the 30 fabrics prepared in the Examples was determined as follows: a 5 centimeter diameter circle of the fabric was cut in half and one of the halves was placed into a teabag (6 cm x 8.5 cm); the weight of the fabric prior to placing in the teabag was recorded; the teabag was sealed and placed in 0.9 % saline solution for 20 35 minutes then centrifuged for three minutes at 1350 revolutions per minute. The weight of the centrifuged teabag was measured and the CRC, in grams per square meter, was determined using the following formula:

40

$$\text{CRC} = (\text{Wt. 2} - \text{Wt. 1} - \text{Blank}) / A$$

Wt. 1 = sample wt.

Wt. 2 = teabag wt after centrifuged

Blank = wt average of two measurements of an empty teabag after centrifuging

45

A = area of sample in square meters  $[(\pi r^2) / 2]$  or 0.000982

The free swell expansion volume (FSEV) is determined by measuring the height (thickness) change, in millimeters, of a compressed web material during hydration. The FSEV of the fabrics indicated in the Examples, were determined as follows: The fabrics were 5 compressed in a Carver Laboratory Press Model #2697 at 7000 pounds of applied load for 48 seconds with the top platen heated to 50°C; a 5 centimeter diameter circle of the fabric was cut from the fabric and the thickness was measured before compression at approximately 4.5 millimeters and after compression at approximately 10 0.67 millimeters using a Fowler Ultra-digit gauge. The weight of the circle was recorded and the circle was placed in a dry sample holder; a single 30 milliliter dose of 0.9% saline was poured on top of the circle; height measurements were taken, with the help of software designed for this purpose, over a 30-minute 15 timeframe every 1.5 seconds. The change in the height of the fabric was measured with a linear variable differential transformer (LVDT, Schaevitz MP-1000) and the data are reported in millimeters (volume) at a chosen time interval.

20 The expansion volume under load (EVUL) is determined by measuring the volume change of a sample as height (thickness) change, in millimeters, of a compressed web material during hydration while under a load. The EVUL is determined in a similar manner as the FSEV except that a 55.93 gram weight (0.5 psi load) is applied to 25 the fabric.

FSEV and EVUL values obtained between 5-30 seconds are useful for the characterization of webs designed for use as acquisition layers while values obtained after 30 minutes are useful for the 30 characterization of webs intended to be used as storage layers. The viscosity was measured in a Brookfield viscometer at a temperature of 20°C at 20 rpm using spindle 02.

Examples

35 The term "superabsorbent polymer particles" used in the Examples means particles of a crosslinked polyacrylic acid having a degree of neutralization of 70 % by mole, said particles were partially neutralized with sodiumhydroxide and surfac-crosslinked. The following acquisition layers, can also be used alone as dual layer.

Example 1

Layered absorbent structure containing an acquisition layer in 45 combination with a storage layer

## (a) Preparation of acquisition layer

An aqueous solution of acrylic acid was prepared by combining 306 grams of acrylic acid, 334 grams of deionized water, 250 grams of 5 a 50 % strength by weight aqueous sodium hydroxide solution, 9.1 grams of SR-9035 available from Sartomer (ethoxylated trimethylolpropane triacrylate), 0.75 grams of Irgacure 2959 (4-(2-hydroxyethoxy)phenyl-(2-hydroxy-propyl)ketone, CAS reg.No. 106797-53-9) from Ciba, 0.45 g of ammonium persulfate and 0.75 10 grams of 2,2'-azobis[2-(2-imidazolin-2-yl)propane]dihydrochloride. The aqueous solution of acrylic acid was purged with nitrogen. Superabsorbent polymer particles were sieved through a 100 mesh screen and the particles passing through having a particle sizes below 150  $\mu\text{m}$  (the mean particle diameter was 80  $\mu\text{m}$ ) 15 were collected. A sprayable blend was prepared by adding 23 grams of said superabsorbent polymer particles to the aqueous acrylic acid solution prepared above under stirring. The resulting sprayable mixture had a viscosity of 53 mPas. This mixture was sprayed using a commercial airless paint spraying apparatus onto both 20 sides of a 0.056  $\text{m}^2$  piece of a commercially available polyester non-woven fabric (HDK-210) having a basis weight of 55 g/ $\text{m}^2$ . The total amount of sprayable mixture retained by the fabric was 10.1 grams (wet load).

25 The fabric was then cured by ultraviolet radiation ("UV") by passing the fabric under a 600 watt/in mercury vapor lamp, using a conveyor operating at 24 feet per minute. The fabric was passed under the lamp twice, turned over and passed under the lamp another two times for a total exposure of 3.75 Joules of UVA 30 radiation per  $\text{cm}^2$  on each side. Following the UV cure, the fabric was dried in an oven for 10 minutes at 150°C. The weight of polymer in the resulting fabric was 82.9 g/ $\text{m}^2$  (60 % by weight of polymer and 40 % by weight of polyester non-woven fabric). The resulting web had a CRC of 7.8 g/g or 1,076 g/ $\text{m}^2$ . The FSEV and EVUL 35 after 5 seconds for the web were 1.9 and 0.8 ml, respectively.

## (b) Preparation of storage layer

An aqueous solution of acrylic acid was prepared by combining 306 40 grams of acrylic acid, 334 grams of deionized water, 250 grams of a 50% strength by weight aqueous sodium hydroxide solution, 1.8 grams of SR-344 available from Sartomer (polyethylene glycol diacrylate 400), 1.15 grams of Darocur 1173 (phenyl-2-hydroxy-2-propyl-ketone) from Merck, 0.15 g of ammonium persulfate and 0.75 45 grams of 2,2'-azobis[2-(2-imidazolin-2-yl)propane]dihydrochloride. The aqueous solution of acrylic acid was purged with nitrogen. Superabsorbent polymer particles were sieved through a 100

mesh screen and the particles passing through having particle sizes below 150  $\mu\text{m}$  (the mean particle diameter was 80  $\mu\text{m}$ ) were collected. A sprayable blend was prepared by adding 23 grams of these superabsorbent polymer particles with particle size below 5 150  $\mu\text{m}$  to the aqueous acrylic acid solution prepared above under stirring. The resulting sprayable mixture had a viscosity of 56 mPas. This mixture was sprayed using a commercial airless paint spraying apparatus onto both sides of a 0.056  $\text{m}^2$  piece of a poly-ester non-woven fabric (HP-15 available from Hobbs Industries) 10 having a basis weight of 80 g/ $\text{m}^2$ . The total amount of sprayable mixture retained by the fabric was 55 grams (wet load). The fabric was then cured by ultraviolet radiation ("UV") and dried as described above. The weight of polymer in the resulting fabric was 453 g/ $\text{m}^2$ . The coated fabric consisted of 85% by weight 15 of polymer and 15 % by weight of polyester non-woven fabric). The resulting coated web had a CRC of 23.0 g/g or 12,259 g/ $\text{m}^2$ . The FSEV and EVUL after 30 minutes for the web were 27 and 18 ml, respectively.

20 (c) Preparation of a layered absorbent core structure

The acquisition and storage layers prepared as described above were laid on top of each other and compressed in a Carver Laboratory Press Model #2697 at 7000 pounds of applied load for 48 seconds with the top platen heated to 50°C. The resulting compressed 25 two layered absorbent composite structure had a thickness of 2.1 mm which was maintained until hydration. The profiled absorbent structure had a CRC of 12,988 g/ $\text{m}^2$  and a FSEV and EVUL after 5 seconds of 3.8 and 2.0 ml, respectively. FSEV and EVUL values after 30 minutes were 34 and 19.5, respectively.

Example 2

Layered absorbent structure containing a lowered pH acquisition 35 layer to control skin pH in combination with a storage layer

(a) Preparation of acquisition layer

An aqueous solution of acrylic acid was prepared by combining 40 368.5 grams of acrylic acid, 441 grams of deionized water, 90.2 grams of 50% strength by weight aqueous sodium hydroxide solution, 9.1 grams of SR-9035 available from Sartomer (ethoxylated trimethylolpropane triacrylate), 0.75 grams of (4-(2-hydroxyethoxy)phenyl-(2-hydroxy-propyl)ketone (Irgacure 2959 from Ciba), 45 0.45 g of ammonium persulfate and 0.75 grams of 2,2'-azo-bis[2-(2-imidazolin-2-yl)propane]dihydrochloride. The aqueous solution of acrylic acid was purged with nitrogen. 12 g of superab-

sorbent polymer particles which were sieved through a 100 mesh screen and had a particle size below 150  $\mu\text{m}$  (mean particle diameter was 45  $\mu\text{m}$ ) were added under stirring to the aqueous acrylic acid solution prepared above. The resulting mixture was sprayable 5 and had a viscosity of 63 mPas. This mixture was sprayed using a commercial airless paint spraying apparatus onto both sides of a 0.056  $\text{m}^2$  piece of a polyester non-woven fabric (HDK-210) having a basis weight of 55 g/ $\text{m}^2$ . The total amount of sprayable mixture retained by the fabric was 17.1 grams (wet load).

10

The fabric was then cured by ultraviolet radiation ("UV") by passing the fabric under a 600 watt/in mercury vapor lamp, using a conveyor operating at 24 feet per minute. The fabric was passed under the lamp twice, turned over and passed under the lamp 15 another two times for a total exposure of 3.75 Joules of UVA radiation per square centimeter on each side. Following the UV cure, the fabric was dried in an oven for 10 minutes at 150°C. The weight of polymer in the resulting coated fabric was 83.7 g/ $\text{m}^2$  (The coated material consisted of 60% by weight of polymer and 40 20 % by weight of fibers in the form of non-woven fabric). The resulting web had a CRC of 4.0 g/g or 335 g/ $\text{m}^2$ . The FSEV and EVUL after 5 seconds for the web were 0.8 and 0.4 ml, respectively. The pH of the surface of the web was 4.5 (measured after the determination of CRC using a Ross Flat Surface Combination pH electrode, Model #8235).

(b) Preparation of storage layer

An aqueous solution of acrylic acid was prepared by combining 306 30 grams of acrylic acid, 334 grams of deionized water, 250 grams of 50% strength by weight aqueous sodium hydroxide solution, 1.8 grams of SR-344 available from Sartomer (polyethylene glycol diacrylate 400), 1.15 grams of Phenyl-2-hydroxy-2-propyl-ketone (Dacrocure 1173 from Merck), 0.15 g of ammonium persulfate and 0.75 35 grams of 2,2'-azobis[2-(2-imidazolin-2-yl)propane]dihydrochloride. The aqueous solution of acrylic acid was purged with nitrogen. A sprayable blend was prepared by adding under stirring 23 grams of superabsorbent polymer particles having a particle size below 150  $\mu\text{m}$  (obtained by sieving superabsorbent particles 40 through a 100 mesh screen, the mean particle diameter was 45  $\mu\text{m}$ ) to the aqueous acrylic acid solution prepared above. The resulting sprayable mixture had a brookfield viscosity of 56 mPas. This mixture was sprayed using a commercial airless paint spraying apparatus onto both sides of a 0.056  $\text{m}^2$  piece of a polyester non-woven fabric (HP-15 available from Hobbs Industries) having a basis 45 weight of 80 g/ $\text{m}^2$ . The total amount of sprayable mixture retained by the fabric was 55 grams (wet load). The fabric was then cured

by ultraviolet radiation ("UV") and dried as described above. The weight of polymer in the resulting coated fabric was 453 g/m<sup>2</sup> (The coated non-woven fabric consisted of 85 % by weight of polymer and 15 % by weight of fibers in the form of non-woven fabric).  
5 The resulting web had a CRC of 23.0 g/g or 12,259 g/m<sup>2</sup>. The FSEV and EVUL after 30 minutes for the web were 27 and 18 ml, respectively.

(c) Preparation of Layered Absorbent Core Structure

10 The acquisition and storage layers prepared as described above were laid on top of each other and compressed in a Carver Laboratory Press Model #2697 at 7000 pounds of applied load for 48 seconds with the top platen heated to 50°C. The resulting compressed 15 two layered absorbent composite structure had a thickness of 2.3 mm which was maintained until hydration. The profiled-absorbent structure had a CRC of 12,509 g/m<sup>2</sup> and a FSEV and EVUL after 5 seconds of 2.7 and 1.0 ml, respectively. FSEV and EVUL values after 30 minutes were 31 and 17.5, respectively.

20 Example 3

Layered absorbent structure containing an acquisition layer in combination with a storage layer and wicking layers

25 (a) Preparation of the acquisition layer

A sprayable blend was prepared as described in Example 1 (a). A polyester non-woven substrate (HP-15) having a weight of 80 g/m<sup>2</sup> 30 and consisting of two thermally bonded 40 g/m<sup>2</sup> layers was used in this Example. A 0.056 m<sup>2</sup> section of this non-woven substrate was carefully separated with gentle tearing to produce two individual sheets with a basis weight of approximately 40 g/m<sup>2</sup>. One of these sheets was coated on both sides with the sprayable mixture by 35 spraying with a commercial airless paint spraying apparatus. The total amount of the sprayable mixture retained by the fabric was 16.5 grams (wet load). The fabric was then cured by ultraviolet radiation ("UV") and dried as described in Example 1. The weight of polymer in the resulting fabric was 135 g/m<sup>2</sup> (the coated fabric 40 consisted of 77 % by weight of polymer and 23 % by weight of polyester fibers). The resulting coated web had a CRC of 8.8 g/g or 1,540 grams per square meter. The FSEV and EVUL after 5 seconds for the web were 1.9 and 1.0ml, respectively.

## (b) Preparation of storage layer

A sprayable blend was prepared as described in Example 1 (b) above and sprayed using a commercial airless paint spraying apparatus onto both sides of the other 0.056 m<sup>2</sup> half sheet described in Example 3 (a), i.e. the sheet obtained by separating the two sheets of the polyester non-woven sample HP-15. The total amount of sprayable mixture retained by the fabric was 16.5 grams (wet load). The fabric was then cured by ultraviolet radiation ("UV") and dried as described in Example 1. The sheet coated on both sides consisted of 85 % by weight of polymer and of 15 % by weight of fibers. The weight of the coating was 227 g/m<sup>2</sup>. The resulting web had a CRC of 23.2 g/g or 6194 grams per square meter. The FSEV and EVUL after 30 minutes for the web were 12.2 and 15 7.0 ml, respectively.

## (c) Preparation of layered absorbent core structure with wicking layers

The acquisition layer of Example 3 (a) and the storage layer of Example 3 (b) were each individually placed in a laboratory pad former and a 125 g/m<sup>2</sup> layer of wood fluff pulp was air-laid onto a single side of each web. The acquisition and storage layers thus prepared were laid on top of each other with the fluff pulp containing side pointing downwards and compressed in a Carver Laboratory Press Model #2697 at 7000 pounds of applied load for 48 seconds with the top and bottom platens heated to 50°C. The resulting compressed four layered absorbent composite structure had a thickness of 3.1 mm which was maintained until hydration. The profiled absorbent structure had a CRC of 8,892 grams per square meter and a FSEV and EVUL after 5 seconds of 2.7 and 1.0 ml, respectively. FSEV and EVUL values after 30 minutes were 15.2 and 8.9 ml, respectively. Wicking and fluid distribution properties were greatly improved when compared to layered absorbent core structures not containing wood fluff layers.

## Examples 4 and 5

The following two commercial diapers were used to prepare the experimental samples described below:

Diaper A: medium, adult brief

Core: homogeneous wood fluff/superabsorbent polymer pad

45 Weight ratio of wood fluff to superabsorbent polymer in core: 38.8 grams: 7.2 grams

## 54

Secondary chemically treated mechanical pulp ("CTMP")  
core, 23.0 grams

Random, pinpointed bonded top sheet

Density of core: 0.08 grams per cubic centimeter

5 Density of secondary CTMP core: 0.09 grams per cubic  
centimeter

Diaper B: medium, baby

10 Core: homogeneous wood fluff/superabsorbent polymer  
pad

Weight ratio of wood fluff to superabsorbent polymer  
in core: 19.7 grams: 10.3 grams

Polypropylene, point bonded top sheet

Density of core: 0.15 grams per cubic centimeter

15

Experimental diapers were prepared by stretching out Diaper A and  
Diaper B and taping them to a laboratory bench top. The experi-  
mental fabric, as indicated in Table 1 below, was used to replace  
both the acquisition layer and wood fluff/superabsorbent polymer  
20 cores in each diaper. A razor knife was used to carefully cut  
along the edges of the top sheet to expose the diaper core. In  
the case of Diaper A, the secondary CTMP core along with the wood  
fluff/ superabsorbent polymer core was removed and replaced with  
the compressed experimental fabric as indicated in Table 1 below.  
25 In the case of Diaper B, the high loft non-woven (basis weight of  
45 g/m<sup>2</sup> ) and the wood fluff/superabsorbent polymer core was re-  
moved and replaced with the compressed experimental fabric as in-  
dicated in Table 1 below.

30 TABLE 1

	Commercial Diaper	Fabric from
Example 4	Diaper A	Example 3
Example 5	Diaper B	Example 3

40 The commercial and experimental diapers were evaluated with a re-  
wet test, which measures the amount of unabsorbed liquid left on  
the surface of a wetted diaper. The rewet values reported in  
Table 2 below, were determined by taping the diaper to a Plexi-  
glas® board or to the counter top, insulting the center of the  
45 diaper with 0.9% saline using a dosing ring (5 centimeter inside  
diameter with 2.5 centimeter height). Diaper A and the absorbent  
structure of Example 4 were insulted with 100 milliliters of the

saline solution. Diaper B and the absorbent structure of Examples 5 were insulted with 80 milliliters of the saline solution. The amount of time needed for the liquid to be absorbed from the surface of the diaper was recorded as the strike through time. After 5 20 minutes a stack of Whatman #3, 9.0 cm diameter filter paper (W1) was weighed and placed on top of the insulted area. A 3620-gram weight was placed on the filter paper for two minutes. The stack of filter paper was re-weighed and the weight recorded (W2). This procedure was repeated two more times, recording the 10 filter paper weight change and the strike through times.

The rewet values were calculated as follows:

$$W2 - W1 = \text{rewet value (grams)}$$

15

The rewet and strikethrough data reported in Table 2, below, are the average values of the results conducted on three separate diapers.

20 TABLE 2

EXAMPLE	REWET (grams)			STRIKETHROUGH TIME (seconds)		
	1 <sup>st</sup> DOSE	2 <sup>nd</sup> DOSE	3 <sup>rd</sup> DOSE	1 <sup>st</sup> DOSE	2 <sup>nd</sup> DOSE	3 <sup>rd</sup> DOSE
Commercial Diaper A	0.8	14	29	15.5	19.7	25.8
Example 4	0.1	2.0	4.0	10.1	12.3	18.5
Commercial Diaper B	0.08	0.15	8.0	7.6	7.6	16.3
Example 5	0.00	0.03	2.0	4.9	5.2	12.1

30

#### Example 6

Acidic acquisition layer, pH = 4.5

35

#### Preparation of Acquisition Layer

An aqueous solution of acrylic acid was prepared by combining 368.5 grams of acrylic acid, 441 grams of deionized water, 90.2 40 grams of 50% w/w sodium hydroxide, 9.1 grams of SR-9035 available from Sartomer (ethoxylated trimethylolpropane triacrylate), 0.75 grams of (4-(2-hydroxyethoxy)phenyl-(2-hydroxy-propyl)ketone (Irgacure 2959 from Ciba), 0.45 g of ammonium persulfate and 0.75 45 grams of 2,2'-azobis[2-(2-imidazolin-2-yl)propane]dihydrochloride. The aqueous solution of acrylic acid was purged with nitrogen. Superabsorbent polymer particles were sieved through a 100 mesh screen and the particles passing through (having particle

sizes below 150  $\mu\text{m}$ , the mean particle diameter was 80  $\mu\text{m}$ ) were collected. A sprayable blend was prepared by adding 12 grams of the collected superabsorbent polymer particles to the aqueous acrylic acid solution prepared above and stirring. The resulting sprayable mixture had a viscosity of 63 mPas. The sprayable mixture was sprayed using a commercial airless paint spraying apparatus onto both sides of a 0.056 square meter piece of a polyester non-woven fabric (HDK-210) having a basis weight of 55 g/m<sup>2</sup>. The total amount of sprayable mixture retained by the fabric was 17.1 grams ("wet load"). The fabric was then cured by ultraviolet radiation ("UV") by passing the fabric under a 600 watt/in mercury vapor lamp, using a conveyor operating at 24 feet per minute. The fabric was passed under the lamp twice, turned over and passed under the lamp another two times for a total exposure of 3.75 Joules of UVA radiation per square centimeter on each side. Following the UV cure, the fabric was dried in an oven for 10 minutes at 150°C. The weight of polymer in the resulting fabric was 83.7 g/m<sup>2</sup> (60 % by weight of polymer and 40 % by weight of fibers). The resulting web had a CRC of 4.0 g/g or 335 grams per square meter. The FSEV and EVUL after 5 seconds for the web were 0.8 and 0.4 ml, respectively. The pH of the surface of the web was measured following the determination of CRC using a Ross Flat Surface Combination pH electrode (Model #8235). The pH was determined to be 4.5.

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#### Example 7

Layered absorbent structure containing a lowered pH and acquisition layer to control skin pH in combination with a storage layer

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##### (a) Preparation of acquisition layer

An acquisition layer was prepared as described in Example 6.

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##### (b) Preparation of storage layer

An aqueous solution of acrylic acid was prepared by combining 306 grams of acrylic acid, 334 grams of deionized water, 250 grams of 50% w/w sodium hydroxide, 1.8 grams of SR-344 available from Sartomer (polyethylene glycol diacrylate 400), 1.15 grams of phenyl-2-hydroxy-2-propyl-ketone (Darocur 1173 from Merck), 0.15 g of ammonium persulfate and 0.75 grams of 2,2'-azobis[2-(2-imidazolin-2-yl)propane]dihydrochloride. The aqueous solution of acrylic acid was purged with nitrogen. Superabsorbent polymer particles were sieved through a 100 mesh screen and the particles passing through (having particle sizes below 150  $\mu\text{m}$ , the mean par-

tive diameter was 80  $\mu\text{m}$ ) were collected. A sprayable blend was prepared by adding 23 grams of the collected superabsorbent polymer particles to the aqueous acrylic acid solution prepared above and stirring. The resulting sprayable mixture had a brookfield viscosity of 56 cps. The sprayable mixture was sprayed using a commercial airless paint spraying apparatus onto both sides of a 0.056 square meter piece of a polyester non-woven fabric (HP-15 available from Hobbs Industries) having a basis weight of 80 grams per square meter. The total amount of sprayable mixture retained by the fabric was 55 grams ("wet load"). The fabric was then cured by ultraviolet radiation ("UV") and dried as described above. The weight of polymer in the resulting coated fabric was 453 g/m<sup>2</sup> (85 % by weight polymer and 15 % by weight fibers). The resulting web had a CRC of 23.0 g/g or 12,259 g/m<sup>2</sup>. The FSEV and EVUL after 30 minutes for the web were 27 and 18 ml, respectively.

(c) Preparation of layered absorbent core structure

The acquisition and storage layers prepared as described above were laid a top one another and compressed in a Carver Laboratory Press Model #2697 at 7000 pounds of applied load for 48 seconds with the top platen heated to 50°C. The resulting compressed two layered absorbent composite structure had a thickness of 2.3 mm which was maintained until hydration. The profiled absorbent structure had a CRC of 12,509 grams per square meter and a FSEV and EVUL after 5 seconds of 2.7 and 1.0 ml, respectively. FSEV and EVUL values after 30 minutes were 31 and 17.5, respectively.

30 Examples 8 - 9

The following two commercial diapers were used to prepare the experimental samples described below:

35 Diaper A: medium, adult brief  
Core:homogeneous wood fluff/superabsorbent polymer pad  
Weight ratio of wood fluff to superabsorbent polymer in core: 38.8 grams: 7.2 grams  
40 Secondary chemically treated mechanical pulp ("CTMP") core, 23.0 grams  
Random, pinpointed bonded top sheet  
Density of core: 0.08 grams per cubic centimeter  
Density of secondary CTMP core: 0.09 grams per cubic centimeter

Diaper B: medium, baby  
 Core: homogeneous wood fluff/superabsorbent polymer pad  
 Weight ratio of wood fluff to superabsorbent polymer in core: 19.7 grams: 10.3 grams  
 5 Polypropylene, point bonded top sheet  
 Density of core: 0.15 grams per cubic centimeter

Experimental diapers were prepared by stretching out Diaper A and 10 Diaper B and taping them to a laboratory bench top. The experimental fabric, as indicated in Table 3 below, was compressed under a 7000 pound load for 48 seconds using a Carver laboratory press with the top platen heated to 50°C. The thickness of the fabric following compression was 0.67 millimeters. An Exacto 15 razor knife was used to carefully cut along the edges of the top sheet to expose the diaper core. In the case of Diaper A, the secondary CTMP core was removed and replaced with the compressed experimental fabric as indicated in Table 3. In the case of Diaper B, the high loft non-woven (basis weight of 45 grams per 20 square meter) was removed and replaced with the compressed experimental fabric as indicated in Table 3.

TABLE 3

25

	Commercial Diaper	Fabric from
Example 8	Diaper A	Example 6
Example 9	Diaper B	Example 6

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The commercial and experimental diapers were evaluated with a rewet test which measures the amount of unabsorbed liquid left on the surface of a wetted diaper. The rewet values reported in 35 Table 4 below, were determined by taping the diaper to a Plexiglas® board or to the counter top, insulting the center of the diaper with 0.9% saline solution using a dosing ring (5 centimeter inside diameter with 2.5 centimeter height). Diaper A and example 8 were insulted with 100 milliliters of the saline. Diaper B and example 9 were insulted with 80 milliliters of the saline. 40 The amount of time needed for the liquid to be absorbed from the surface of the diaper was recorded as the strike through time. After 20 minutes a stack of Whatman #3, 9.0 cm diameter filter paper (W1) was weighed and placed on top of the insulted area. A 3620- gram weight was placed on the filter paper for two minutes. 45 The stack of filter paper was reweighed and the weight recorded (W2). This procedure was repeated two more times, recording the filter paper weight change and the strike through times. The pH

of the top sheet, the transfer layer (if any) and the core were measured after the experiment was complete using a Ross Flat Surface Epoxy Combination pH Model #8235 electrode; the data are reported in Table 5. The rewet values were calculated as follows:

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$W_2 - W_1$  = rewet value (grams)

The rewet and strikethrough data reported in Table 4 are the average values of the results conducted on three separate diapers.

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TABLE 4

EXAMPLE	REWET (grams)			REWET (grams) STRIKETHROUGH TIME (seconds)		
	1 <sup>st</sup> DOSE	2 <sup>nd</sup> DOSE	3 <sup>rd</sup> DOSE	1 <sup>st</sup> DOSE	2 <sup>nd</sup> DOSE	3 <sup>rd</sup> DOSE
Commercial Diaper A	0.8	14	29	15.5	19.7	25.8
Example 8	1.1	8.1	18.4	4.4	4.8	4.9
Commercial Diaper B	0.08	0.15	8.0	7.6	7.6	16.3
Example 9	0.56	0.65	0.78	3.3	4.0	4.3

25

TABLE 5

pH MEASUREMENTS OF DIAPER SURFACES			
EXAMPLE	TOPSHEET	ACQUISITION LAYER	FLUFF/SAP CORE
Commercial Diaper A	5.3	5.4	5.8
Example 8	4.7	4.4	5.1
Commercial Diaper B	5.3	5.4	5.8
Example 9	4.5	4.5	5.1

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Example 10

Double sided coated profiled absorbent structure containing an acquisition layer in combination with preparation of a storage layer

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(a) Preparation of storage layer

An aqueous solution of acrylic acid was prepared by combining 306 grams of acrylic acid, 334 grams of deionized water, 250 grams of 50% w/w sodium hydroxide, 1.8 grams of SR-344 available from Sartomer (polyethylene glycol diacrylate 400), 1.15 grams of phe-

nyl-2-hydroxy-2-propyl-ketone (Darocur 1173 from Merck), 0.15 g of ammonium persulfate and 0.75 grams of 2,2'-azobis[2-(2-imidazolin-2-yl)propane]dihydrochloride. The aqueous solution of acrylic acid was purged with nitrogen. Superabsorbent polymer particles were sieved through a 100 mesh screen and the particles passing through (having particle sizes below 150  $\mu\text{m}$ , the mean particle diameter was 45  $\mu\text{m}$ ) were collected. A sprayable blend was prepared by adding 45 grams of the collected superabsorbent polymer particles to the aqueous acrylic acid solution prepared above and stirring. The resulting sprayable mixture had a brookfield viscosity of 56 mPas. The sprayable mixture was sprayed using a commercial airless paint spraying apparatus onto one side of a 0.056 square meter piece of a polyester non-woven fabric (HP-15 available from Hobbs Industries) having a basis weight of 80 g/m<sup>2</sup>. The total amount of sprayable mixture retained by the fabric was 55 grams ("wet load"). The fabric was then cured by ultraviolet radiation ("UV") by passing the fabric under a 600 watt/in mercury vapor lamp, using a conveyor operating at 24 feet per minute. The fabric was passed under the lamp twice, turned over and passed under the lamp another two times for a total exposure of 3.75 Joules of UVA radiation per square centimeter on each side. Following the UV cure, the fabric was dried in an oven for 10 minutes at 150°C. The weight of polymer in the resulting fabric was 453 grams per square meter (85 % by weight of polymer and 15 % by weight of fibers). The resulting web had a CRC of 23.0 g/g or 12,259 g/m<sup>2</sup>. The FSEV and EVUL after 30 minutes for the web were 27 and 18 ml, respectively.

(b) Preparation of acquisition layer

An aqueous solution of acrylic acid was prepared by combining 306 grams of acrylic acid, 334 grams of deionized water, 250 grams of 50% w/w sodium hydroxide, 9.1 grams of SR-9035 available from Sartomer (ethoxylated trimethylolpropane triacrylate), 0.75 grams of (4-(2-hydroxyethoxy)phenyl-(2-hydroxy-propyl)ketone (Irgacure 2959 from Ciba), 0.45 g of ammonium persulfate and 0.75 grams of 2,2'-azobis[2-(2-imidazolin-2-yl)propane]dihydrochloride. The aqueous solution of acrylic acid was purged with nitrogen. Superabsorbent polymer particles were sieved through a 100 mesh screen and the particles passing through (having particle sizes below 150  $\mu\text{m}$ , the mean particle diameter was 45  $\mu\text{m}$ ) were collected. A sprayable blend was prepared by adding 22.5 grams of the collected superabsorbent polymer particles to the aqueous acrylic acid solution prepared above and stirring. The resulting sprayable mixture had a Brookfield viscosity of 63 mPas. The sprayable mixture was sprayed using a commercial airless paint spraying apparatus onto the uncoated side of the storage layer prepared as

described above. The total amount of sprayable mixture retained by the fabric was 17.1 grams ("wet load"). The fabric was then cured by ultraviolet radiation ("UV") and dried as described above. The weight of additional superabsorbent polymer added to the resulting fabric was 83.7 grams per square meter (60 % by weight of polymer and 40 % by weight of fibers). The profiled absorbent structure had a CRC of 12,988 grams per square meter and a FSEV and EVUL after 5 seconds of 3.8 and 2.0 ml, respectively. FSEV and EVUL values after 30 minutes were 34 and 19.5, respectively.

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#### Example 11

Double sided coated profiled absorbent structure containing a lowered pH acquisition layer in combination with a storage layer

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##### (a) Preparation of storage layer

An aqueous solution of acrylic acid was prepared by combining 306 grams of acrylic acid, 334 grams of deionized water, 250 grams of 20 50% w/w sodium hydroxide, 1.8 grams of SR-344 available from Sartomer (polyethylene glycol diacrylate 400), 1.15 grams of phenyl-2-hydroxy-2-propyl-ketone (Darocur 1173 from Merck), 0.15 g of ammonium persulfate and 0.75 grams of 2,2'-azobis[2-(2-imidazolin-2-yl)propane]dihydrochloride. The aqueous solution of acrylic acid was purged with nitrogen. Superabsorbent polymer particles were sieved through a 100 mesh screen and the particles passing through (having particle sizes below 150  $\mu\text{m}$ , the mean particle diameter was 80  $\mu\text{m}$ ) were collected. A sprayable blend was prepared by adding 45 grams of the collected superabsorbent polymer particles to the aqueous acrylic acid solution prepared above and stirring. The resulting sprayable mixture had a Brookfield viscosity of 56 cps. The sprayable mixture was sprayed using a commercial airless paint spraying apparatus onto one side of a 0.056 square meter piece of a polyester non-woven fabric (HP-15 30 available from Hobbs Industries) having a basis weight of 80 g/m<sup>2</sup>. The total amount of sprayable mixture retained by the fabric was 55 grams ("wet load"). The fabric was then cured by ultraviolet radiation ("UV") by passing the fabric under a 600 watt/in mercury vapor lamp, using a conveyor operating at 24 feet per minute. The fabric was passed under the lamp twice, turned over and passed under the lamp another two times for a total exposure of 3.75 Joules of UVA radiation per square centimeter on each side. Following the UV cure, the fabric was dried in an oven for 10 40 minutes at 150°C. The weight of polymer in the resulting fabric was 453 g/m<sup>2</sup> (85 % by weight of polymer and 15 % by weight of fibers). The resulting web had a CRC of 23.0 g/g or 12,259 g/m<sup>2</sup>. The

FSEV and EVUL after 30 minutes for the web were 27 and 18 ml, respectively.

(b) Preparation of acquisition layer

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An aqueous solution of acrylic acid was prepared by combining 368.5 grams of acrylic acid, 441 grams of deionized water, 90.2 grams of 50% w/w sodium hydroxide, 9.1 grams of SR-9035 available from Sartomer (ethoxylated trimethylolpropane triacrylate), 0.75 10 grams of (4-(2-hydroxyethoxy)phenyl-(2-hydroxy-propyl)ketone (Irgacure 2959 from Ciba), 0.45 g of ammonium persulfate and 0.75 grams of 2,2'-azobis[2-(2-imidazolin-2-yl)propane]dihydrochloride. The aqueous solution of acrylic acid was purged with nitrogen. Superabsorbent polymer particles were sieved through a 100 15 mesh screen and the particles passing through (having particle sizes below 150  $\mu\text{m}$ , the mean particle diameter was 80  $\mu\text{m}$ ) were collected. A sprayable blend was prepared by adding 22.5 grams of the collected superabsorbent polymer particles to the aqueous acrylic acid solution prepared above and stirring. The resulting 20 sprayable mixture had a brookfield viscosity of 63 mPas. The sprayable mixture was sprayed using a commercial airless paint spraying apparatus onto the uncoated side of the web prepared as described above. The total amount of sprayable mixture retained by the fabric was 17.1 grams ("wet load"). The fabric was then cured 25 by ultraviolet radiation ("UV") and dried as described above. The weight of additional superabsorbent polymer added to the resulting fabric was 83.7 grams per square meter (60 % by weight of polymer and 40 % by weight of fibers). The profiled absorbent structure had a CRC of 12,509 g/m<sup>2</sup> and a FSEV and EVUL after 5 30 seconds of 2.7 and 1.0 ml, respectively. FSEV and EVUL values after 30 minutes were 31 and 17.5, respectively. The pH of the surface of the web was measured following the determination of CRC using a Ross Flat Surface Combination pH electrode (Model #8235). The pH of the acquisition side of the was determined to be 4.5. 35 The pH of the storage layer side of the web was 5.9.

Example 12

Commercial adult brief incorporating an acquisition/storage layer 40 core supplement

The following commercial diapers was used to prepare the experimental sample described below:

Diaper A: medium, adult brief  
Core:homogeneous wood fluff/superabsorbent polymer  
pad  
Weight ratio of wood fluff to superabsorbent polymer  
5 in core: 38.8 grams: 7.2 grams  
Secondary chemically treated mechanical pulp ("CTMP")  
core, 23.0 grams  
Random, pinpointed bonded top sheet  
Density of core: 0.08 grams per cubic centimeter  
10 Density of secondary CTMP core: 0.09 grams per cubic  
centimeter

Experimental diapers were prepared by stretching out Diaper A and taping them to a laboratory bench top. The experimental fabric, 15 as prepared in Example 11, was compressed under a 7000 pound load for 48 seconds using a Carver laboratory press with both platens heated to 50°C. The thickness of the fabric following compression was 1.35 millimeters. An Exacto razor knife was used to carefully cut along the edges of the top sheet to expose the diaper core. 20 The secondary CTMP core was removed and replaced with the compressed experimental fabric.

The commercial and experimental diapers were evaluated with a re-wet test which measures the amount of unabsorbed liquid left on 25 the surface of a wetted diaper. The rewet values reported in Table 6 below, were determined by taping the diaper to a Plexi-glas® board or to the counter top, insulting the center of the diaper with 0.9% saline solution using a dosing ring (5 centimeter inside diameter with 2.5 centimeter height). Diaper A and the 30 absorbent member of Example 3 were insulted with 100 milliliters of the saline. The amount of time needed for the liquid to be absorbed from the surface of the diaper was recorded as the strike through time. After 20 minutes a stack of Whatman #3, 9.0 cm diameter filter paper (W1) was weighed and placed on top of the insulted area. A 3620- gram weight was placed on the filter paper 35 for two minutes. The stack of filter paper was re-weighed and the weight recorded (W2). This procedure was repeated two more times, recording the filter paper weight change and the strike through times. The pH of the top sheet, the transfer layer (if any) and 40 the core were measured after the experiment was complete using a Ross Flat Surface Epoxy Combination pH Model #8235 electrode. The data are reported in Table 7, below:

The rewet values were calculated as follows:

45

W2 -W1 = rewet value (grams)

The rewet and strikethrough data reported in Table 6, below, are the average values of the results conducted on three separate diapers.

5 TABLE 6

EXAMPLE	REWET (grams)			STRIKETHROUGH TIME (seconds)		
	1 <sup>st</sup> DOSE	2 <sup>nd</sup> DOSE	3 <sup>rd</sup> DOSE	1 <sup>st</sup> DOSE	2 <sup>nd</sup> DOSE	3 <sup>rd</sup> DOSE
Commercial Diaper A	0.8	14	29	15.5	19.7	25.8
Absorbent member of Example 12	0.1	8.4	18.2	8.6	11.2	18.7

TABLE 7

EXAMPLE	TOPSHEET	ACQUISITION		FLUFF/SAP CORE
		LAYER		
Commercial Diaper A	5.3	5.4		5.8
Absorbent member of Example 12	4.9	5.0		5.5

20 Other layers which can be used as acquisition or dual layers are described in the following examples.

## Example 13

25 Acquisition layer containing an odor control agent

An aqueous solution of acrylic acid was prepared by combining 306 grams of acrylic acid, 334 grams of deionized water, 250 30 grams of 50 % w/w sodium hydroxide, 9.1 grams of SR-9035 available from Sartomer (ethoxylated trimethylolpropane triacrylate), 0.75 grams of 4-(2-hydroxyethoxy)phenyl-(2-hydroxy-2-propyl)ketone (Irgacure 2959 from Ciba) 0.45 g of ammonium persulfate and 0.75 grams of 2,2'-azobis[2-(2-imidazolin-2-yl)propane]dihydrochloride. To the monomer mixture 15.3 g of technical grade beta-cyclodextrin BW7 (available from Wacker Biochemicals) was added as an odor control agent. The aqueous solution of acrylic acid was purged with nitrogen. Superabsorbent polymer particles were sieved through a 100 mesh screen and the particles passing 35 through (having particle sizes below 150 microns, mean particle diameter was 80  $\mu\text{m}$ ) were collected. A sprayable blend was prepared by adding 23 grams of the collected superabsorbent polymer particles to the aqueous acrylic acid solution prepared above and stirring. The resulting sprayable mixture had a brookfield viscosity of 53 mPas. the sprayable mixture was sprayed using a commercial airless paint spraying apparatus onto both sides of a 40 0.056  $\text{m}^2$  piece of a polyester non-woven fabric (HDK-210) having a 45

basis weight of 55 grams per square meter. The total amount of sprayable mixture retained by the fabric was 10.1 grams (wet load). The fabric was then cured by ultraviolet radiation ("UV") by passing the fabric under a 600 watt/in mercury vapor lamp, using a conveyor operating at 24 feet per minute. The fabric was passed under the lamp twice, turned over and passed under the lamp another two times for a total exposure of 3.75 Joules of UVA radiation per square centimeter on each side. Following the UV cure, the fabric was dried in an oven for 10 minutes at 150°C. The weight of polymer in the resulting fabric was 82.9 g/m<sup>2</sup> (60 wt%). The resulting web had a CRC of 7.8 g/g or 1.076 g/m<sup>2</sup>. The FSEV and EVUL after 5 seconds for the web were 1.9 and 0.8 ml, respectively.

#### 15 Example 14

An acquisition layer was prepared identical to that in Example 13 except that 54.0 g of Bioshield AM 5001 (available from Bioshield Technologies, Inc. in Norcross Ga.) was substituted for beta-cyclodextrin. The active ingredient in Bioshield AM 500 I is 3-(trimethoxysilyl)propyldimethyloctadecyl ammonium chloride.

#### Example 15

25 An acquisition layer was prepared identical to that in Example 13 except that no odor control agent was added to the sprayable blend prior to polymerization. A solution was prepared by dissolving 0.10 g of Irgaguard B 1000 (Triclosan, available from Ciba) in 100 ml of a 50/50 (v/v) propylene glycol/water solution. 30 The solution was placed in a thin layer chromatography atomizing sprayer and 3.0 g of the solution was applied to each side of the web. The coated web was then dried in an oven for 10 minutes at 150°C.

#### 35 Example 16

A 0.056 m<sup>2</sup> piece of a polyester non-woven fabric (HDK-210) having a basis weight of 55 g/m<sup>2</sup> was coated with the Irgaguard B 1000 solution as described in Example 15. The resulting Triclosan 40 treated web was then used to prepare an acquisition layer as described in Example 13.

## Example 17

Layered absorbent structure containing an acquisition layer in combination with a storage layer incorporating odor control

## 5 agents

## (a) Preparation of Acquisition layer

An acquisition layer was prepared as described in Example 14.

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## (b) Preparation of Storage layer

An aqueous solution of acrylic acid was prepared by combining 306 grams of acrylic acid, 334 grams of deionized water, 15 250 grams of 50 % w/w sodium hydroxide, 1.8 grams of SR-344, available from Sartomer (polyethylene glycol diacrylate 400), 1.15 grams of, the mean particle diameter was 45  $\mu\text{m}$  (Darocur 1173 from Merck), 0.15 g of ammonium persulfate and 0.75 grams of 2,2'-azobis[2-(2-imidazolin-2-yl)propane]dihydrochloride. To the monomer mixture 54.0 g of Bioshield AM 500 l (available from Bioshield Technologies, Inc. in Norcross Ga.) was added as an odor control agent. The aqueous solution of acrylic acid was purged with nitrogen. Superabsorbent polymer particles were sieved through a 100 mesh screen and the particles passing through (having particle sizes below 150 microns, the mean particle diameter was 45  $\mu\text{m}$ ) were collected. A sprayable blend was prepared by adding 23 grams of the collected superabsorbent polymer particles to the aqueous acrylic acid solution prepared above and stirring. The resulting sprayable mixture had a brookfield viscosity of 56 mPas. The sprayable mixture was sprayed using a commercial airless paint spraying apparatus onto both sides of a 0.056 square meter piece of a polyester non-woven fabric (HP-15 available from Hobbs Industries) having a basis weight of 80 g/m<sup>2</sup>. The 20 25 30 35 40 total amount of sprayable mixture retained by the fabric was 55 grams ("wet load"). The fabric was then cured by ultraviolet radiation ("UV") and dried as described above. The weight of polymer in the resulting fabric was 453 grams per square meter (85 wt%). The resulting web had a CRC of 23.0 g/g or 12.259 grams per square meter. The FSEV and EVUL after 30 minutes for the web were 27 and 18 ml, respectively.

## (c) Preparation of Layered Absorbent Core Structure

45 The acquisition and storage layers prepared as described above were laid a top one another and compressed in a Carver Laboratory Press Model #2697 at 7000 pounds of applied load.

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for 48 seconds with the top platen heated to 50°C. The resulting compressed two layered absorbent composite structure had a thickness of 2.1 mm which was maintained until hydration. The profiled absorbent structure had a CRC of 12.988 grams per square meter and a FSEV and EVUL after 5 seconds of 3.8 and 2.0 ml, respectively. FSEV and EVUL values after 30 minutes were 34 and 19.5, respectively.

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## Claims

1. Use of a layer obtainable by a process comprising
  - 5 (a) forming a sprayable blend comprising one or more superabsorbent forming monomers superabsorbent polymer particles water, and one or more initiators,
  - 10 (b) applying said sprayable blend on a fibrous web; and
  - 15 (c) subjecting said fibrous web to conditions under which the superabsorbent forming monomer with polymerize, as a storage layer for aqueous fluids.
2. Use of a layer according to claim 1 in hygenic products.
- 20 3. A storage layer as described in claim 1 with a CRC equal or greater 9000 g/m<sup>2</sup>, preferable greater 10000 g/m<sup>2</sup>, more preferable greater 12000 g/m<sup>2</sup>.
- 25 4. A storage layer as described in claim 1 or of claim 3 which contains an odor control agent or agent with skin care effect or combination of both.
5. A storage layer of one of the claims 3 or 4 wherein the sprayable blend has a viscosity at least 20 mPas.
- 30 6. A storage layer of one of the claims 3 to 5 which is part of a double sided coated fibrous web wherein the other side is an acquisition layer.
- 35 7. A storage layer as claimed in one of the claims 3 to 6, wherein the blend formed in step (a) further comprises a cross-linking agent.
- 40 8. A storage layer as claimed in one of the claims 3 to 7, wherein the blend formed in step (a) further comprises a softening agent.
- 45 9. A storage layer as claimed in one of the claims 3 to 8, wherein the blend formed in step (a) contains at least one odor control means or an agent with skin care effect.

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10. A storage layer as claimed in one of the claims 3 to 9, wherein the superabsorbent polymer particles of the blend formed in step (a) contain at least one odor control means or an agent with skin care effect.

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11. A storage layer as claimed in one of the 3 to 10, wherein the storage layer has a pH of from 2.0 to 7.5, preferably from 4.0 to 6.5.

10 12. A storage layer as claimed in one of the claims 3 to 11, wherein the superabsorbent particles consist of mixed bed ion exchange superabsorbent polymers or multidomain composites.

15 13. An absorbent article comprising a storage layer as claimed in one of the claims 3 to 12.

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## INTERNATIONAL SEARCH REPORT

Internal	Application No
PCT/EP 02/11516	

A. CLASSIFICATION OF SUBJECT MATTER		
IPC 7	A61L15/60	A61L15/46
A61F13/15		

According to International Patent Classification (IPC) or to both national classification and IPC
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B. FIELDS SEARCHED
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Minimum documentation searched (classification system followed by classification symbols)
IPC 7 A61L A61F

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched
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Electronic data base consulted during the international search (name of data base and, where practical, search terms used)
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EPO-Internal, WPI Data, PAJ
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C. DOCUMENTS CONSIDERED TO BE RELEVANT
--

Category	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	WO 01 56625 A (BASF CORP) 9 August 2001 (2001-08-09) cited in the application claims; examples -----	1-13
E	WO 02 094328 A (BASF AG ;ENGELHARDT FRIEDRICH (DE); WHITMORE DARRYL L (US)) 28 November 2002 (2002-11-28) the whole document -----	1-13
E	WO 02 094329 A (BASF AG ;ENGELHARDT FRIEDRICH (DE); WHITMORE DARRYL L (US)) 28 November 2002 (2002-11-28) the whole document ----- -/-	1-13

<input checked="" type="checkbox"/> Further documents are listed in the continuation of box C.
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<input checked="" type="checkbox"/> Patent family members are listed in annex.
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Date of the actual completion of the international search
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Date of mailing of the international search report
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5 February 2003
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## INTERNATIONAL SEARCH REPORT

Internal	Application No
PCT/EP	02/11516

## C.(Continuation) DOCUMENTS CONSIDERED TO BE RELEVANT

Category	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
A	EP 0 290 814 A (MITSUBISHI PETROCHEMICAL CO ;UNI CHARM CORP (JP)) 17 November 1988 (1988-11-17) claims; examples ----	1-13
A	US 5 217 445 A (MOORE DANNY R ET AL) 8 June 1993 (1993-06-08) claims; examples I-II ----	1-13
A	WO 92 11830 A (PROCTER & GAMBLE) 23 July 1992 (1992-07-23) claims ----	1-13
A	WO 98 26808 A (TRINH TOAN ;PROCTER & GAMBLE (US)) 25 June 1998 (1998-06-25) claims ----	1-13

**INTERNATIONAL SEARCH REPORT**

Information on patent family members

Intern: Application No  
PCT/cr 02/11516

Patent document cited in search report		Publication date		Patent family member(s)	Publication date
WO 0156625	A	09-08-2001	US	6417425 B1	09-07-2002
			BR	0107973 A	05-11-2002
			CA	2333212 A1	01-08-2001
			EP	1251886 A2	30-10-2002
			WO	0156625 A2	09-08-2001
WO 02094328	A	28-11-2002	WO	02094328 A2	28-11-2002
WO 02094329	A	28-11-2002	WO	02094329 A1	28-11-2002
EP 0290814	A	17-11-1988	JP	2036403 C	28-03-1996
			JP	7064896 B	12-07-1995
			JP	63260906 A	27-10-1988
			AU	608825 B2	18-04-1991
			AU	1471088 A	20-10-1988
			CA	1305098 A1	14-07-1992
			DE	3875793 D1	17-12-1992
			DE	3875793 T2	08-04-1993
			EP	0290814 A2	17-11-1988
			ES	2036612 T3	01-06-1993
			KR	9512335 B1	17-10-1995
			US	4892754 A	09-01-1990
US 5217445	A	08-06-1993	AT	153527 T	15-06-1997
			AU	659406 B2	18-05-1995
			AU	7149991 A	21-08-1991
			BR	9105943 A	27-10-1992
			CA	2068540 C	04-06-1996
			CN	1053543 A , B	07-08-1991
			DE	69126299 D1	03-07-1997
			DE	69126299 T2	25-09-1997
			DK	512001 T3	30-06-1997
			EP	0512001 A1	11-11-1992
			ES	2103796 T3	01-10-1997
			FI	923340 A	22-07-1992
			GR	3023575 T3	29-08-1997
			HK	1006527 A1	05-03-1999
			IE	910225 A1	31-07-1991
			JP	5503445 T	10-06-1993
			NZ	236855 A	25-02-1994
			SG	54165 A1	16-11-1998
			TR	27618 A	13-06-1995
			WO	9111165 A1	08-08-1991
			PT	96543 A , B	15-10-1991
WO 9211830	A	23-07-1992	AT	119382 T	15-03-1995
			AU	665937 B2	25-01-1996
			AU	9167191 A	17-08-1992
			BR	9107262 A	19-04-1994
			CA	2098322 A1	04-07-1992
			CN	1066776 A , B	09-12-1992
			CZ	9301335 A3	16-03-1994
			DE	69108059 D1	13-04-1995
			DE	69108059 T2	07-09-1995
			DK	565606 T3	22-05-1995
			EP	0565606 A1	20-10-1993
			ES	2069416 T3	01-05-1995
			FI	933067 A	02-09-1993

**INTERNATIONAL SEARCH REPORT**

Information on patent family members

Intern: Application No

PCT/EP 02/11516

Patent document cited in search report	Publication date		Patent family member(s)	Publication date
WO 9211830	A		GR 3015375 T3	30-06-1995
			HK 94296 A	07-06-1996
			HU 67763 A2	28-04-1995
			HU 217875 B	28-04-2000
			IE 920010 A1	15-07-1992
			JP 6503983 T	12-05-1994
			KR 231311 B1	15-11-1999
			MX 9200019 A1	01-08-1992
			NO 932372 A	03-09-1993
			NZ 241204 A	24-03-1997
			NZ 270804 A	24-10-1997
			NZ 328138 A	25-02-1999
			PL 168901 B1	31-05-1996
			PT 99976 A , B	31-05-1994
			RU 2127573 C1	20-03-1999
			SK 68393 A3	11-05-1994
			TR 28310-A	09-04-1996
			WO 9211830 A2	23-07-1992
			US 5439458 A	08-08-1995
			US 5849003 A	15-12-1998
			US 5460622 A	24-10-1995
			US 5486167 A	23-01-1996
			US 5658269 A	19-08-1997
			US 5702382 A	30-12-1997
			US 5611790 A	18-03-1997
			US 5713884 A	03-02-1998
			US 5674212 A	07-10-1997
			US 5683375 A	04-11-1997
			US 6059764 A	09-05-2000
			US 6287288 B1	11-09-2001
			US 5824004 A	20-10-1998
			US 5304161 A	19-04-1994
WO 9826808	A	25-06-1998	AU 739247 B2	04-10-2001
			AU 5597298 A	15-07-1998
			EP 0946209 A2	06-10-1999
			JP 2000505692 T	16-05-2000
			WO 9826808 A2	25-06-1998

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